For each of the following thermodynamic terms in the first list, choose the appropriate description chosen from the second list.

- 1. Control volume
- 2. Property
- 3. Intensive property
- 4. State
- 5. Equilibrium state
- 6. Process
- 7. Quasi-equilibrium process
- 8. Path
- A. Whatever is under study
- B. A property whose value for an overall system is the sum of its values for the parts into which the system is divided
- C. Everything external to the system
- D. A property whose value is independent of the size or extent of a system
- E. A fixed quantity of matter
- F. Distinguishes the system from its surroundings
- G. Series of states for a process
- H. Slow process where the system is always infinitesimally close to equilibrium at each state along the path
- I. A macroscopic characteristic of a system to which a numerical value can be assigned at a given time without knowledge of the previous behavior of the system
- J. A region of space through which mass may flow
- K. Change of system from one equilibrium state to another
- L. Condition where when a system is isolated from its surroundings, there are no observable changes in properties over time
- M. The condition of a system as described by its properties

SOLUTION:

- 1. Control volume: J. A region of space through which mass may flow
- 2. Property: I. A macroscopic characteristic of a system to which a numerical value can be assigned at a given time without knowledge of the previous behavior of the system
- 3. Intensive property: D. A property whose value is independent of the size or extent of a system
- 4. State: M. The condition of a system as described by its properties.
- 5. Equilibrium state: L. Condition where when a system is isolated from its surroundings, there are no observable changes in properties over time.
- 6. Process: K. Change of system from one equilibrium state to another
- 7. Quasi-equilibrium process: H. Slow process where the system is always infinitesimally close to equilibrium at each state along the path
- 8. Path: G. Series of states for a process.