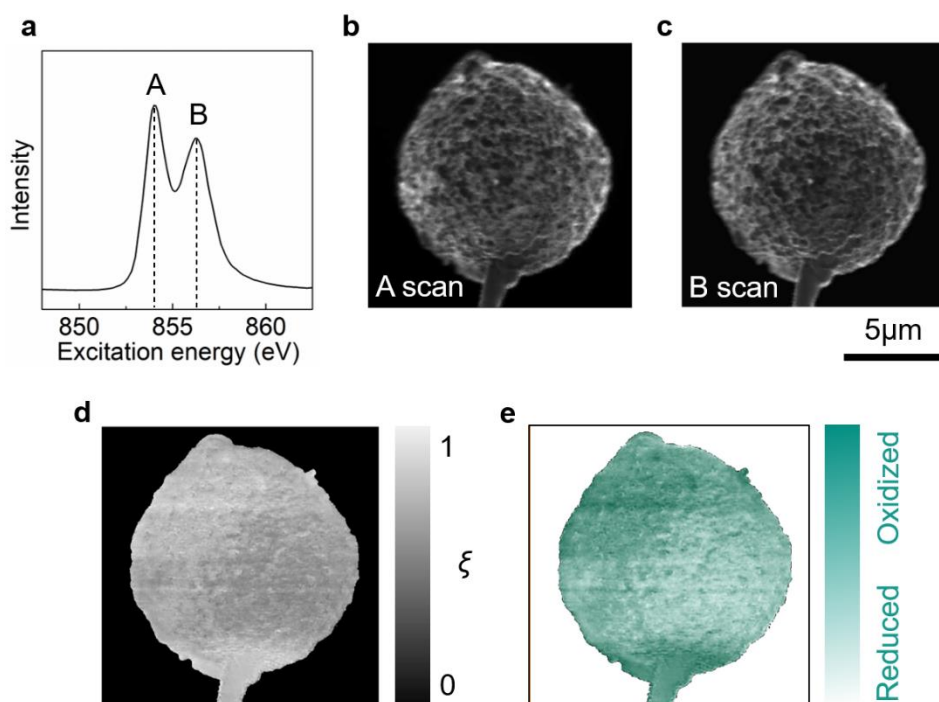


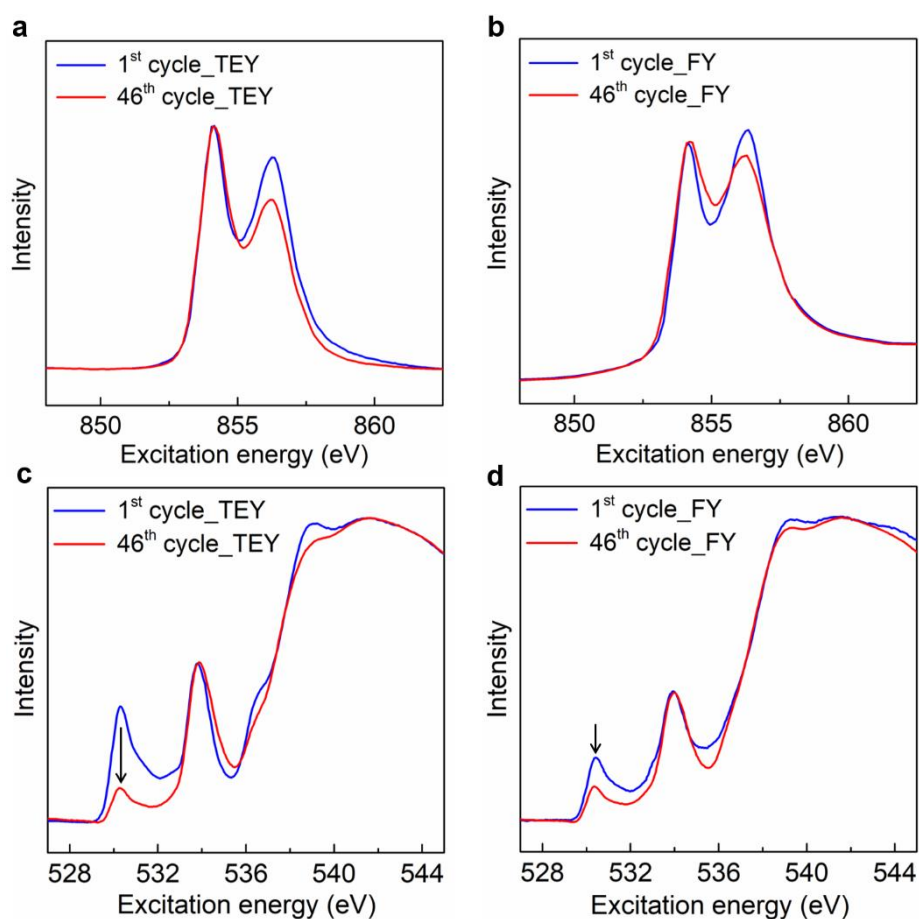
Supplementary Information for

**Mutual modulation between surface chemistry and bulk  
microstructure within secondary particles of Ni-rich layered oxides**

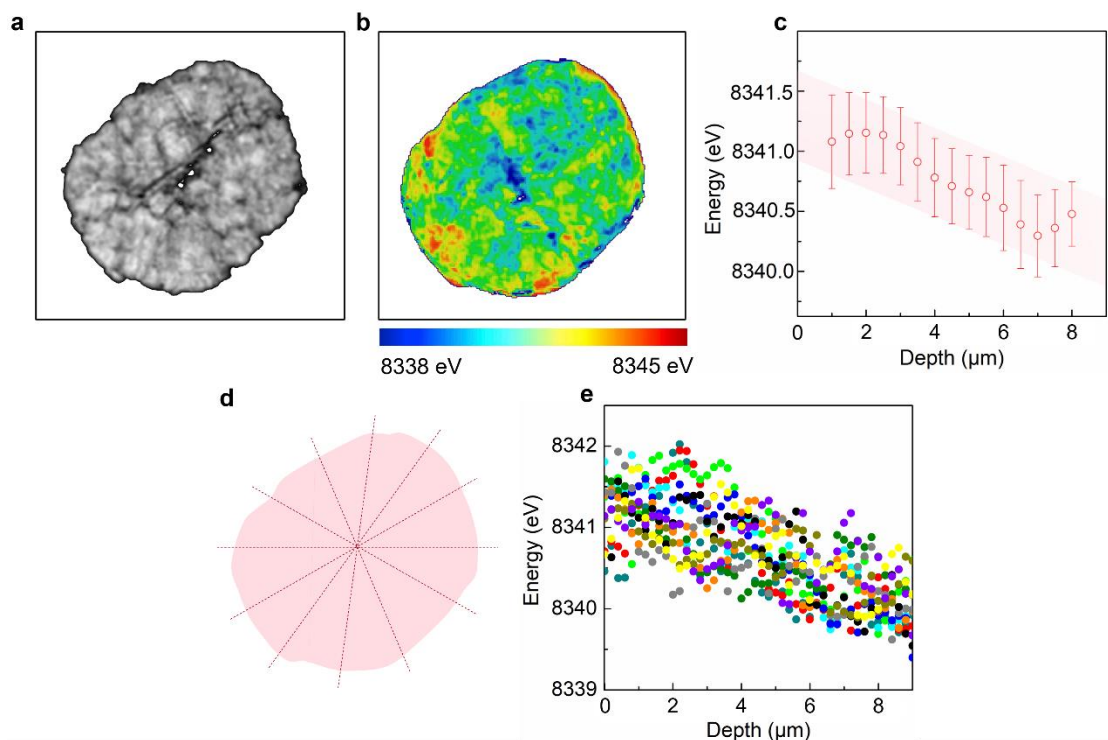
*Li et al.*



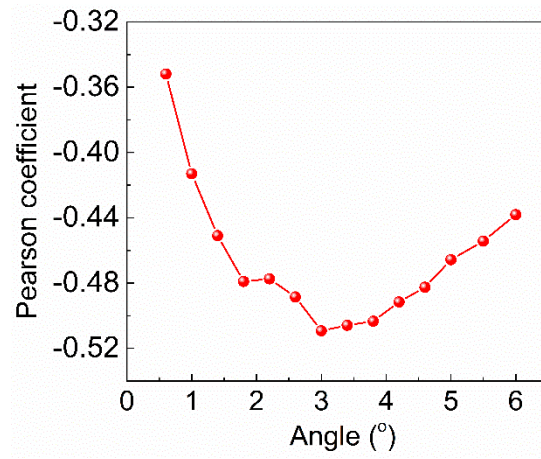
**Supplementary Figure 1. Schematic illustration of using scanning soft x-ray nanoprobes to probe the surface Ni valence heterogeneity.** **a**, Averaged soft x-ray absorption spectra of the charged NMC811 electrode over the Ni  $L_3$ -edge in the TEY mode. **A scan** (**b**) and **B scan** (**c**) are scanning soft x-ray nanoprobes TEY signal collected with incoming x-rays energies of 854.0 eV and 856.2 eV, respectively. **d**, TEY intensity ratio ( $\xi$ ) is calculated using the following equation pixel by pixel:  $\xi_{(x,y)} = \frac{I_{856.2}^{\text{TEY}}}{I_{854.0}^{\text{TEY}} + I_{856.2}^{\text{TEY}}}$ , in which  $(x, y)$  is the position of each pixel,  $I$  is the TEY intensity at selected incoming energy. **e**, Map of Ni valence state, which was proportional to TEY intensity ratio in (**d**).



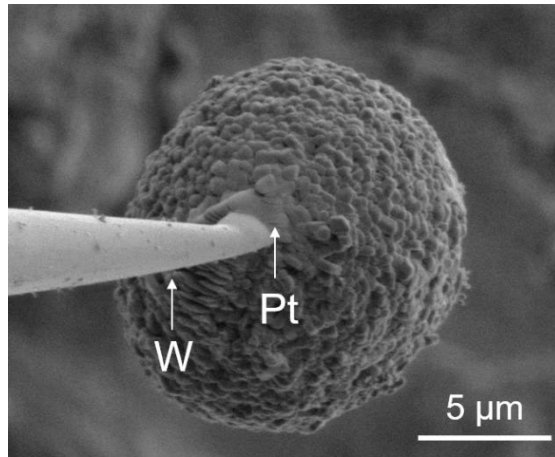
**Supplementary Figure 2. XAS spectra of the charged NMC811 electrode.** NMC811 electrode measured at the Ni  $L_3$ -edge and the O  $K$ -edge via the TEY mode (**a**, **c**) and the FY mode (**b**, **d**). The electrodes recovered after the 1<sup>st</sup> cycle (blue) and the 46<sup>th</sup> cycle (red) are compared to highlight the prolonged electrochemical cycling induced chemical degradation. The black arrows in (**c**) and (**d**) denotes the suppression of the pre-edge peak at excitation energy of  $\sim 530.3$  eV in oxygen  $K$ -edge spectra upon electrochemical cycling.



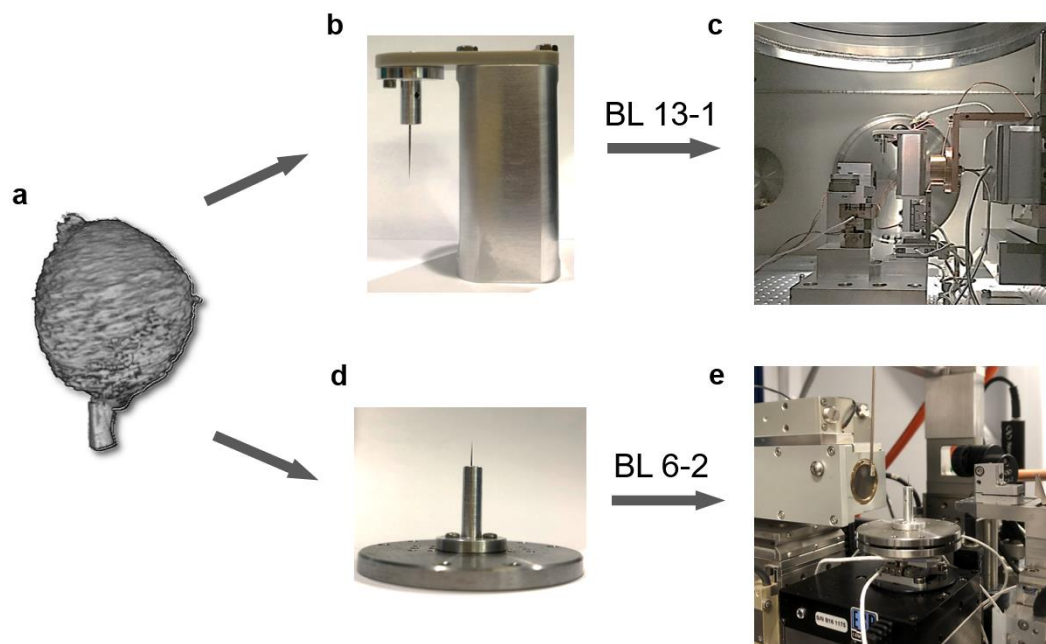
**Supplementary Figure 3. Chemical complexity within a charged NMC811 secondary particle.** **a, b,** The xy-slice through the center of the particle (**a**) and corresponding Ni K-edge energy map (**b**). **c,** The depth profile of the edge energy distribution over the particle. The near-surface regions are generally more oxidized compare to the particle core. **d,** Schematic of the selected lines for plotting the line profiles in (**e**). **e,** Edge energy distribution from the surface to the center over different lines.



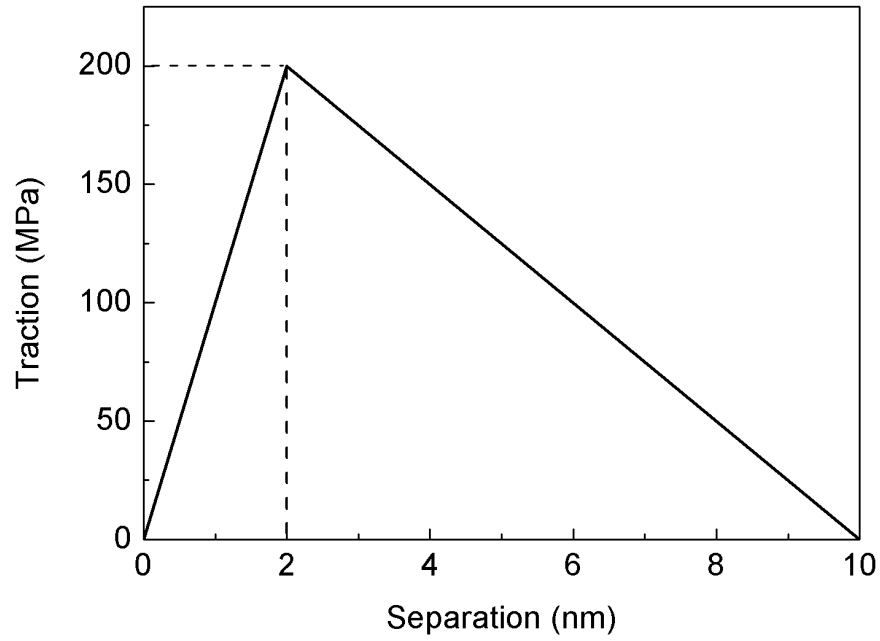
**Supplementary Figure 4. The correlation between the surface mapping data and the bulk porosity with different opening angle for the cone-shape regions.**



**Supplementary Figure 5. SEM image of single NMC811 secondary particle mounted on the W needle with Pt welding.**



**Supplementary Figure 6. Optical photographs of sample holders and beamlines.** a, 3D visualization of single NMC811 secondary particle. Optical photographs of the sample holders for BL13-1 (b) and 6-2c (d). Optical photographs of the sample holders installed on the BL13-1 (c) and 6-2c (e).



**Supplementary Figure 7. The traction-separation law of the cohesive zone model.**

The area under the traction-separation curve represents the fracture toughness.



**Supplementary Table 1. Parameters used in finite element modeling.**

Parameter	Symbol	Value
Particle radius	$r$	$5 \times 10^{-6}$ m
Elastic constants	$C_{11}$	259 GPa
	$C_{12}$	107 GPa
	$C_{13}$	75 GPa
	$C_{33}$	194 GPa
	$C_{44}$	59 GPa
Li diffusivity along $ab$ plane	$D_{ab}$	$7 \times 10^{-15}$ m <sup>2</sup> s <sup>-1</sup>
Li diffusivity along $c$ direction	$D_c$	$7 \times 10^{-16}$ m <sup>2</sup> s <sup>-1</sup>
Li diffusivity in surface passivation layer	$D_L$	$7 \times 10^{-17}$ m <sup>2</sup> s <sup>-1</sup>
Maximum Li concentration	$C_{\max}$	63887 mol m <sup>-3</sup>
Tensile strength	$t_n$	200 MPa
Fracture toughness	$G_c$	1 J m <sup>-2</sup>
Elastic modulus of surface passivation layer	$E$	190 GPa
Poisson's ratio of surface passivation layer	$\nu$	0.35