



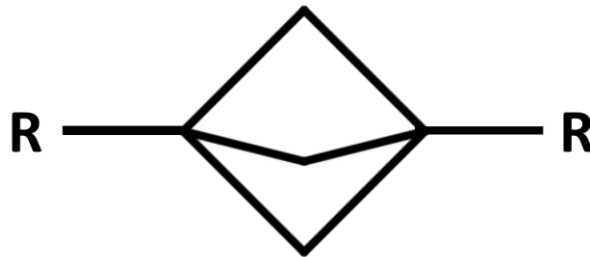
BICYCLO[1.1.1]PENTANE – A RISING PROCESS SAFETY CHALLENGE

CONNOR BARRETT AND JEFFREY SPERRY

DECEMBER 7, 2021

INTRODUCTION

- Bicyclo[1.1.1]pentane (BCP) is a small, 3-dimensional, strained ring structure
- BCPs have found increased use as a bio-isostere in the pharmaceutical industry
- Publications have emerged describing methods for making BCPs and methods for implanting the structure within molecules



Iridium-Catalyzed Enantioselective Synthesis of α -Chiral Bicyclo[1.1.1]pentanes by 1,3-Difunctionalization of [1.1.1]Propellane

Songjie Yu, Changcheng Jing, Adam Noble, and Varinder K. Aggarwal*


doi.org/10.1021/acs.orglett.0c02017

Direct catalytic asymmetric synthesis of α -chiral bicyclo[1.1.1]pentanes

Marie L. J. Wong, Alistair J. Sterling, James J. Mousseau, Fernanda Duarte  & Edward A. Anderson 

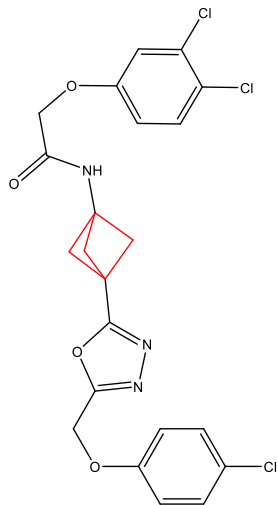
doi.org/10.1038/s41467-021-21936-4

1,2-Difunctionalized bicyclo[1.1.1]pentanes: Long-sought-after mimetics for *ortho/meta*-substituted arenes

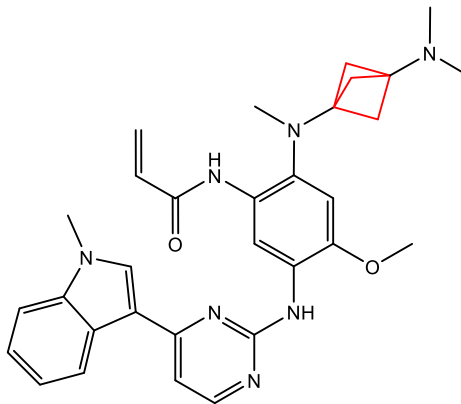
Jin-Xin Zhao, Yu-Xuan Chang, Chi He, Benjamin J. Burke, Michael R. Collins, Matthew Del Bel, Jeff Elleraas, Gary M. Gallego, T. Patrick Montgomery, James J. Mousseau, Sajiv K. Nair, Matthew A. Perry, Jillian E. Spangler, Julien C. Vantourout, and  Phil S. Baran

doi.org/10.1073/pnas.2108881118

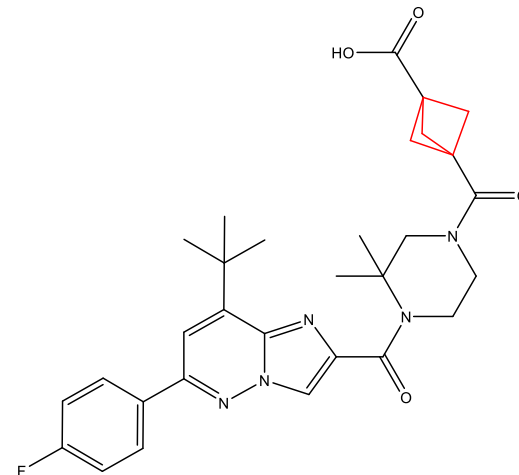
PHARMACEUTICAL UTILIZATION OF THE BCP SCAFFOLD



Calico Life Sciences
WO 2017/193030 A1



Kalyra Pharmaceuticals
WO 2017/205459 A1





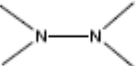
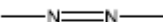
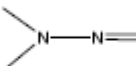
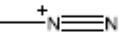

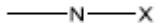
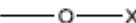
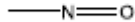
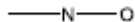
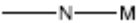
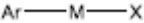



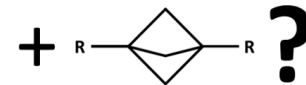
Vertex Pharmaceuticals
WO 2015/048245 A1

HIGH ENERGY FUNCTIONAL GROUPS (HEFG)

- A HEFG list is a valuable guide for flagging potentially energetic compounds
- Testing at earlier stages allows potential issues to be identified before larger scale reactions are performed
- HEFG lists are NOT exhaustive. The list needs to be a living document

Is bicyclo[1.1.1]pentane a HEFG?

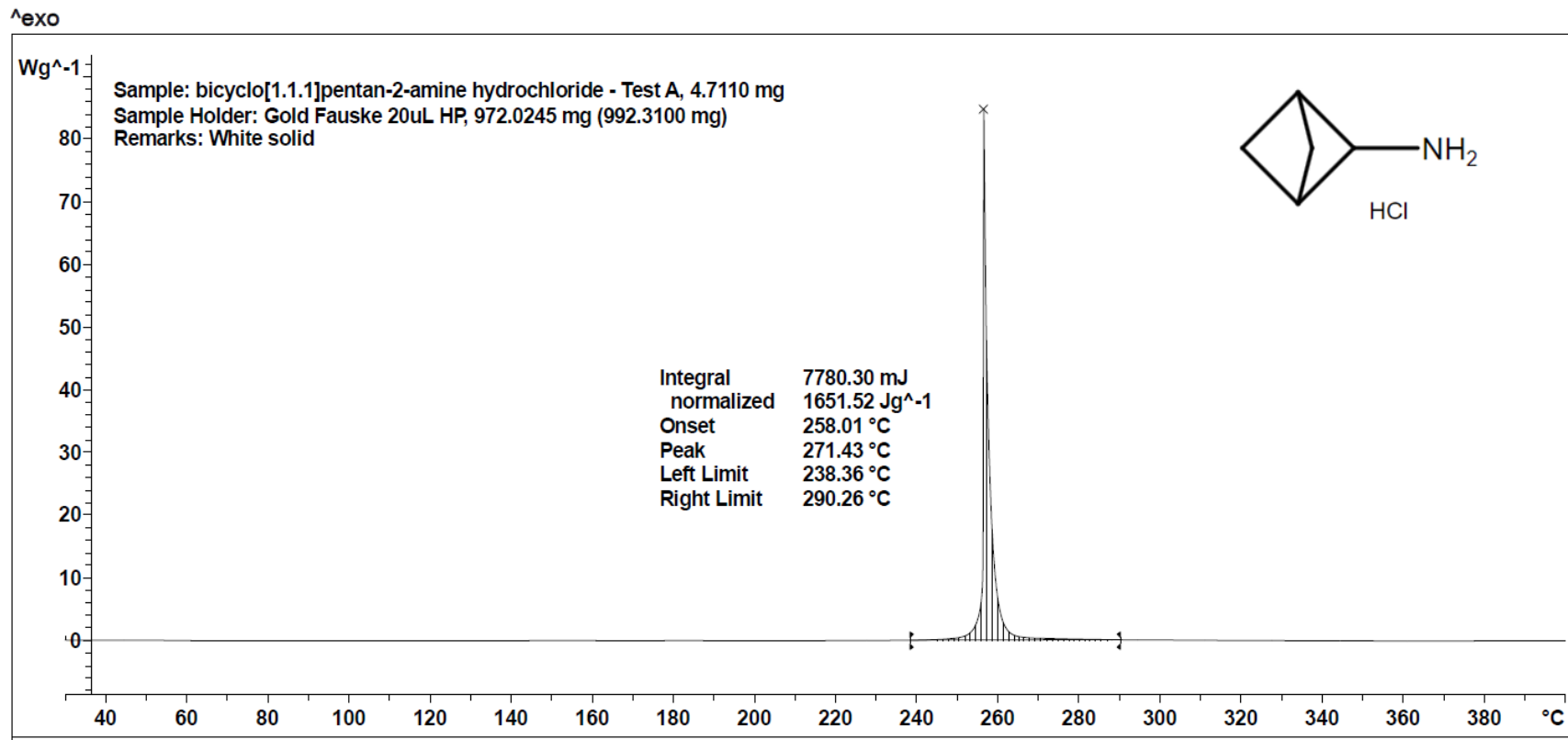
	
	
	
	
	
	
	
	
	



DIFFERENTIAL SCANNING CALORIMETRY (DSC)

- DSC testing offers rapid screening of materials for thermal decomposition hazards
- The speed and low-sample requirement have made the DSC a common starting point in process safety studies
- DSC testing of a readily available BCP showed a high-energy decomposition but more testing was needed to study the thermal stability of BCPs.



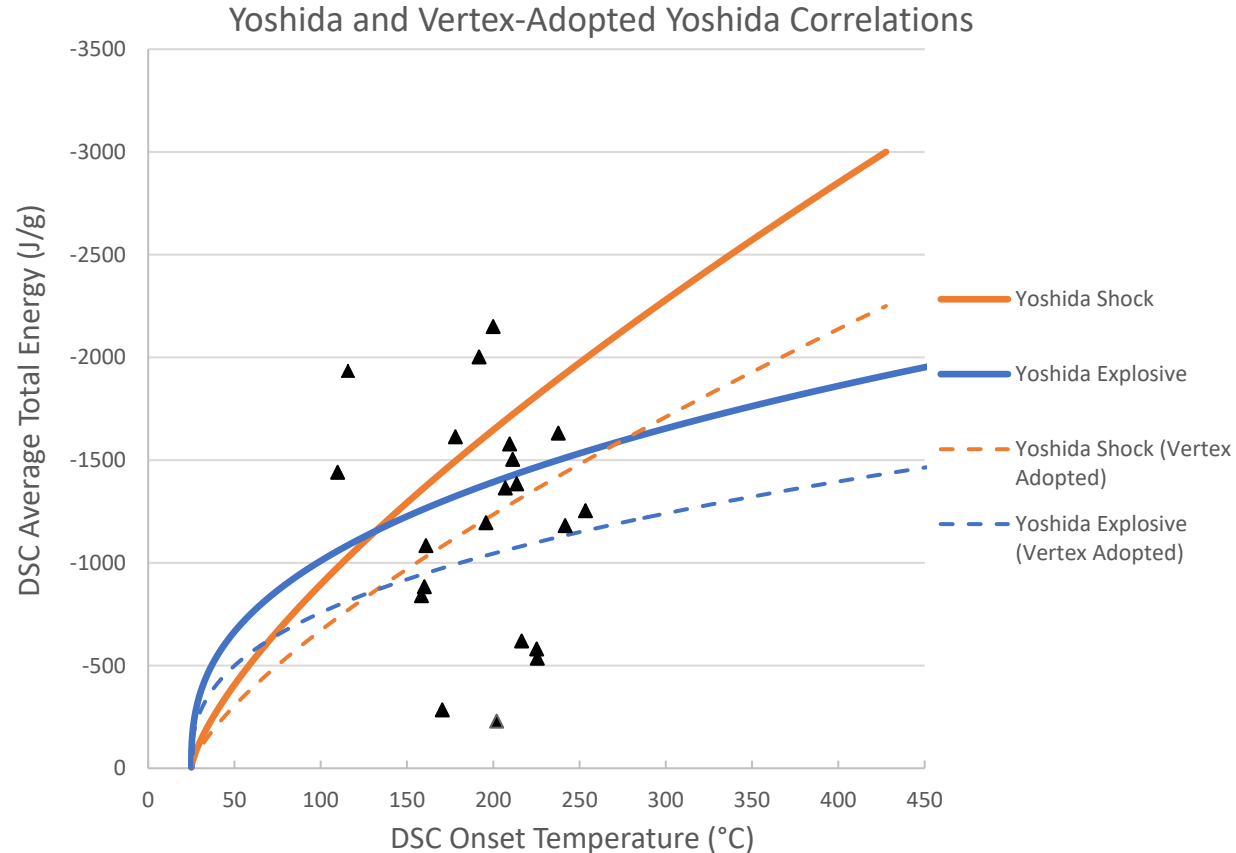


Lab: METTLER

STAR® SW 16.20

YOSHIDA CORRELATION

- The Yoshida Curve is used to predict impact sensitivity and the potential to propagate an explosion
- Several of the first BCP compounds tested flagged on the Vertex-modified Yoshida Curve
- Testing was expanded to readily available commercial BCP compounds
- 14 of the compounds tested flagged explosive, 10 of those flagged shock sensitive



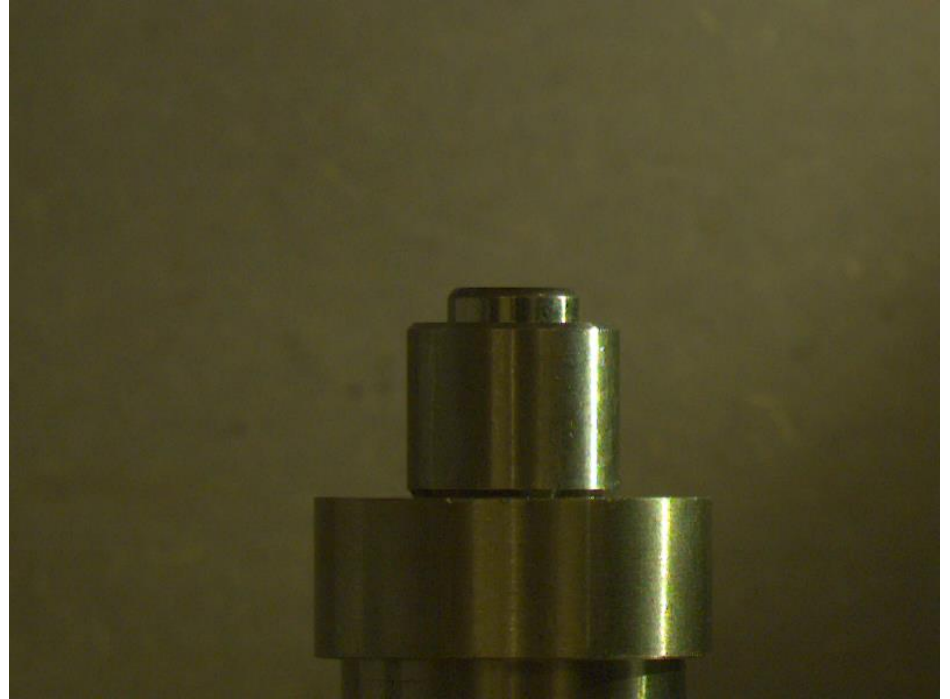
IMPACT SENSITIVITY TESTING

- BAM Fallhammer apparatus was used for testing
- The BAM Fallhammer was designed for the screening of potential high explosives prior to shipment
- While most materials tested are not explosive, the test still offers information on sensitivity to degradation or decomposition without detonation



BCP'S ARE SENSITIVE TO IMPACTS

- Impact sensitivity testing was performed on available samples that flagged on the Vertex-modified Yoshida Curve
- 6 of 10 BCPs tested showed sensitivity to decomposition from impacts. 3 of those samples decomposed with smoke
- No flames, audible reactions, or detonations were detected

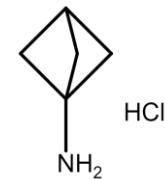


OREOS METHOD

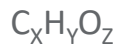
- The OREOS Method helps characterize potentially explosive materials by assigning point values for various properties. The resulting number offers guidance on testing that is either required or recommended
- The Method takes into account the Oxygen Balance, Rule of 6, Explosive Functional Groups, Onset Temperature, and the Scale.

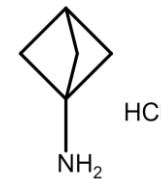
		Points			
		1	2	4	8
Oxygen Balance Hazard			Low	Med	High
Rule of 6 calculation			Pass		Fail
Explosive Functional Group?		No			Yes
Onset temperature		>300	200-300	125-200	<125
Scale		1mg to 5g	5g to 100g	101g to 500g	>500g
O.R.E.O.S. Total:					
Points:	Low Hazard	Medium Hazard		High Hazard	
	7 to 17	18 to 27		28 to 40	
Testing:	No additional testing is required. ARC testing is recommended to understand pressure generation.	ARC testing is required . UN TS1 testing is recommended.		ARC testing is required . UN TS1 testing is required . Alternatives are encouraged.	

BCP-1-AMINE HCL OREOS



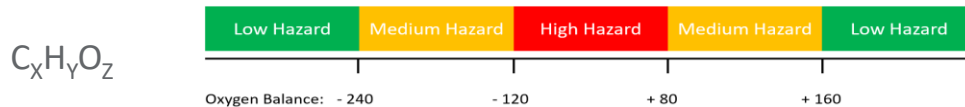
$$\text{Oxygen Balance} = \frac{\left[-1600\left(2X + \frac{Y}{2} - Z\right)\right]}{MW}$$

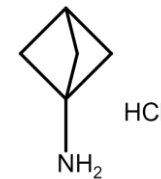




BCP-1-AMINE HCL OREOS

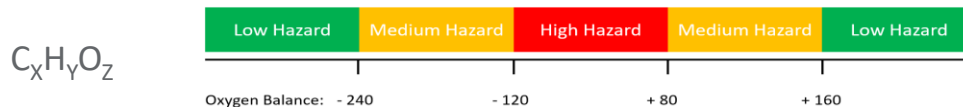
$$\text{Oxygen Balance} = \frac{\left[-1600\left(2X + \frac{Y}{2} - Z\right)\right]}{MW} = \frac{\left[-1600\left(2(5) + \frac{(10)}{2} - 0\right)\right]}{119.59} = -201 = \text{Medium (4 pts)}$$



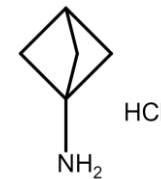


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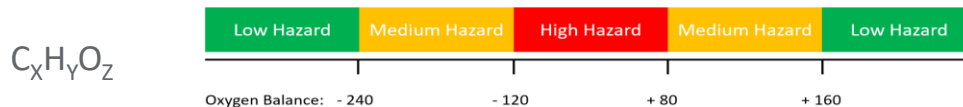


Rule of 6 – No explosive functional groups = **Pass (2 pts)**



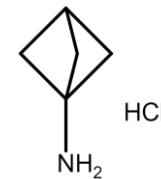
BCP-1-AMINE HCL OREOS

$$\text{Oxygen Balance} = \frac{\left[-1600\left(2X + \frac{Y}{2} - Z\right)\right]}{MW} = \frac{\left[-1600\left(2(5) + \frac{(10)}{2} - 0\right)\right]}{119.59} = -201 = \text{Medium (4 pts)}$$



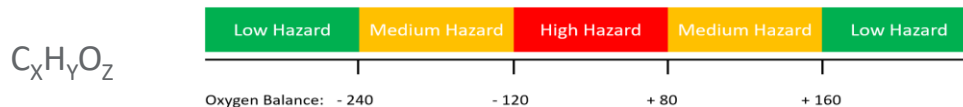
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ExFG – No explosive functional groups = **No (1 pt)**



BCP-1-AMINE HCL OREOS

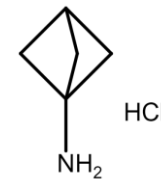
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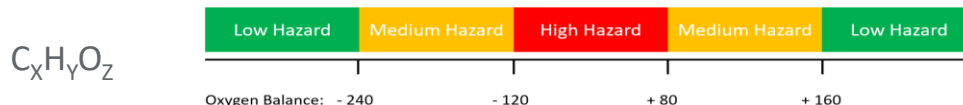
ExFG – No explosive functional groups = **No (1 pt)**

Onset Temperature = 178°C = **Medium (4 pts)**



BCP-1-AMINE HCL OREOS

$$\text{Oxygen Balance} = \frac{\left[-1600 \left(2X + \frac{Y}{2} - Z \right) \right]}{MW} = \frac{\left[-1600 \left(2(5) + \frac{(10)}{2} - 0 \right) \right]}{119.59} = -201 = \text{Medium (4 pts)}$$



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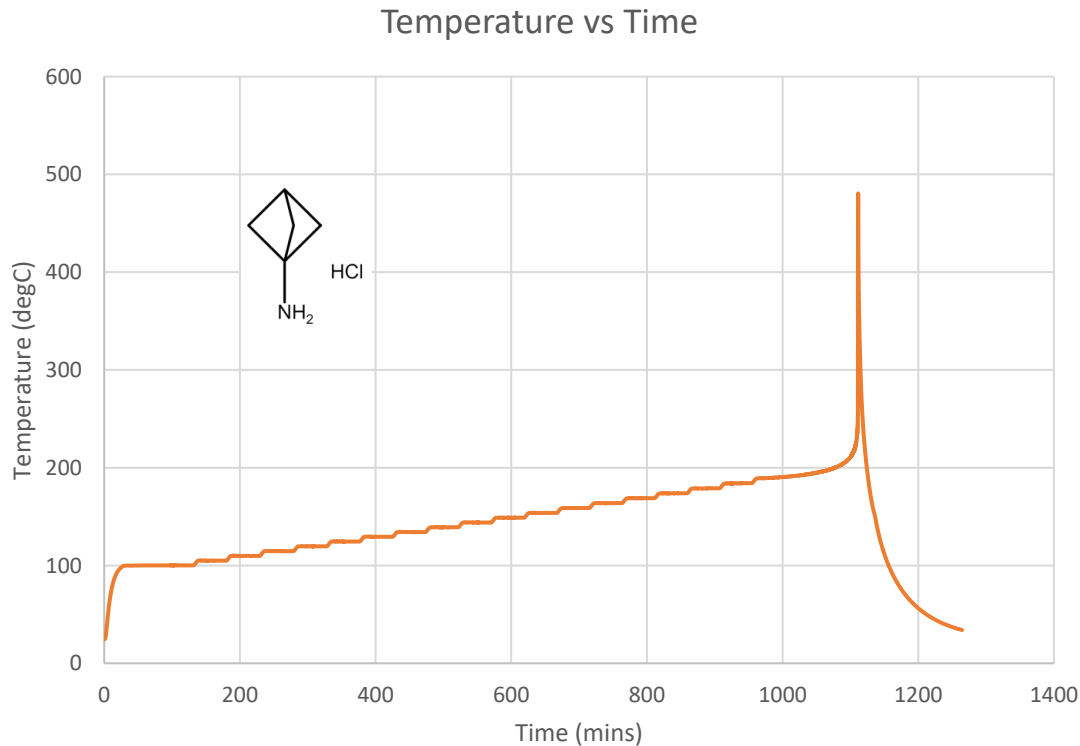
ExFG – No explosive functional groups = **No (1 pt)**

Onset Temperature = 178°C = **Medium (4 pts)**

Scale	<5g	5g to 100g	100g to 500g	>500g
Points	1	2	4	8
Total	12	13	15	19
	Low Hazard	Low Hazard	Low Hazard	Medium Hazard

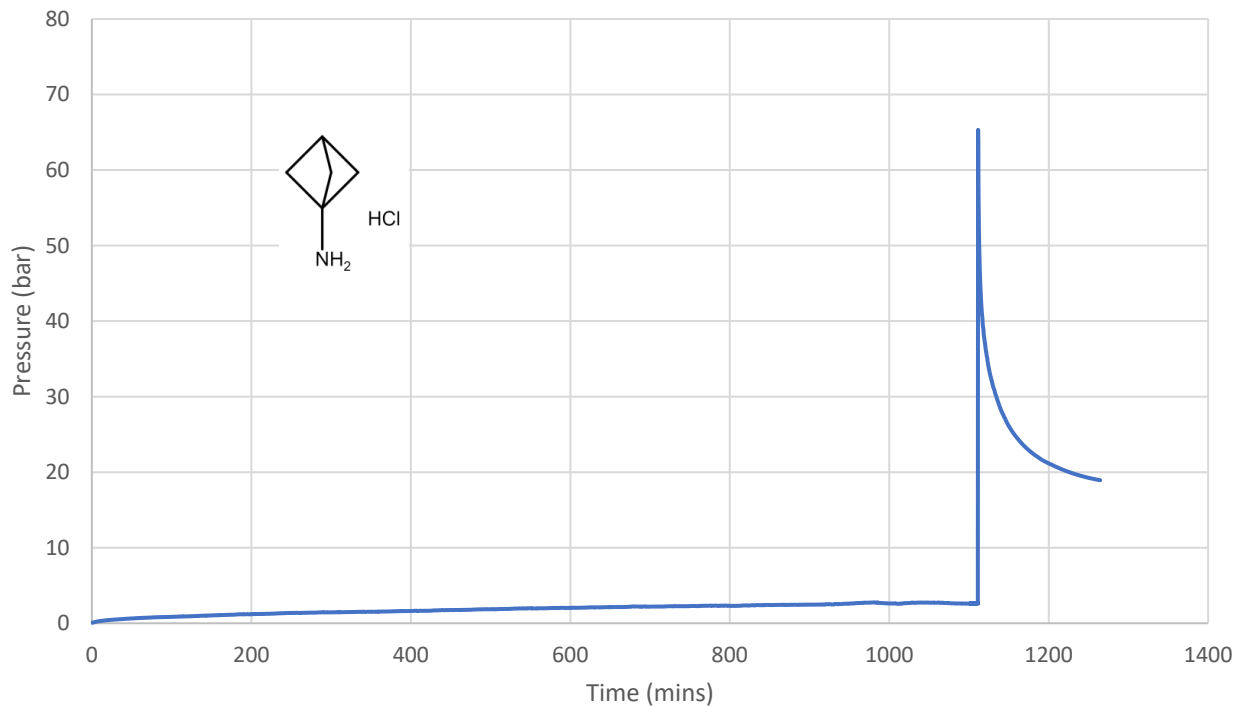
ACCELERATING RATE CALORIMETRY (PHITEC-1)

- ARC testing was performed to get a better idea of the onset temperatures and to generate data on pressure rise rates and/or non-condensable gas generation
- Each of the 5 compounds tested showed rapid temperature and pressure rise resulting in a loss of adiabaticity
- Each compound produced a moderate amount of non-condensable gas with 6 – 18 bar remaining after cooldown



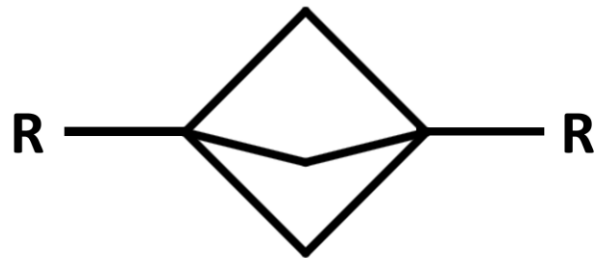
ACCELERATING RATE CALORIMETRY (PHITEC-1)

Pressure vs Time



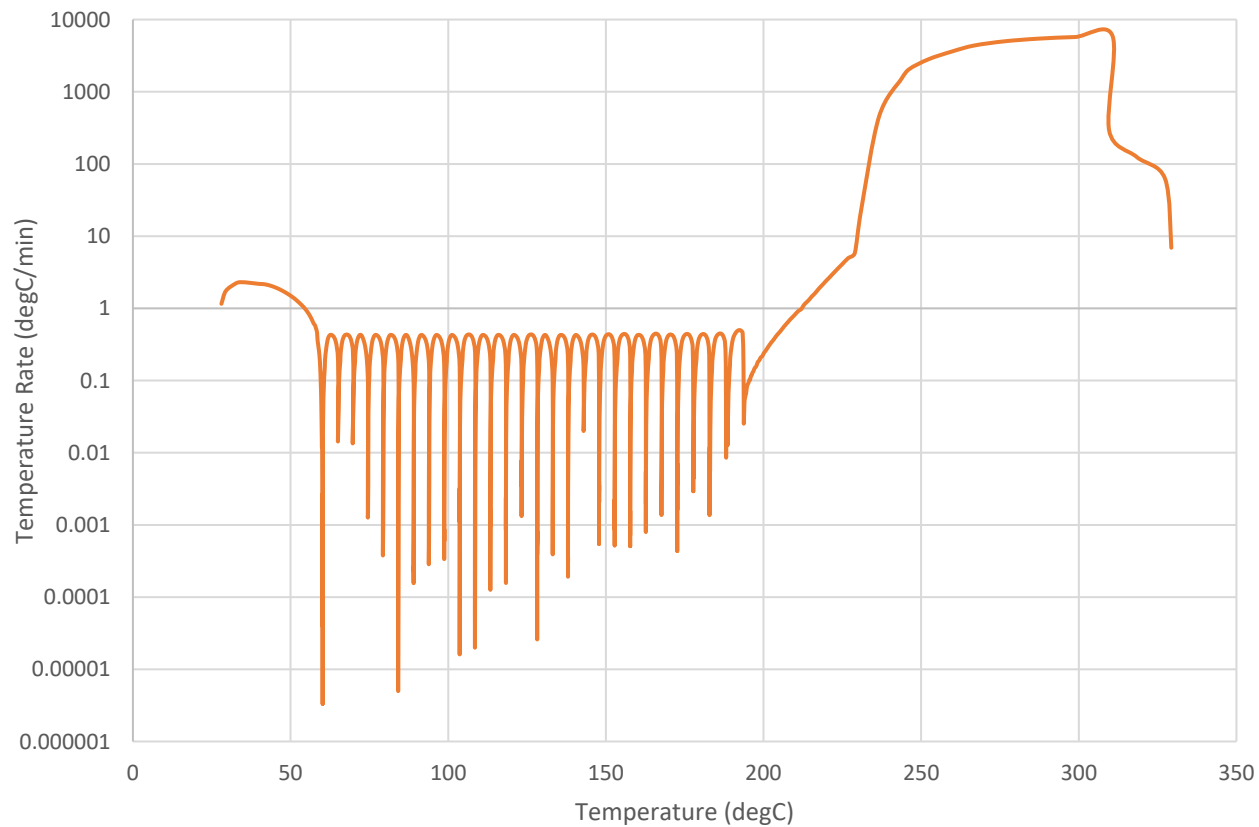
CONCLUSION

- The BCP structure is high energy and caution is needed when using compounds containing BCPs, especially low molecular weight building blocks – “test what you don’t know”
- Having a testing workflow to triage potentially energetic compounds is critical. Methods like the OREOS method help guide testing

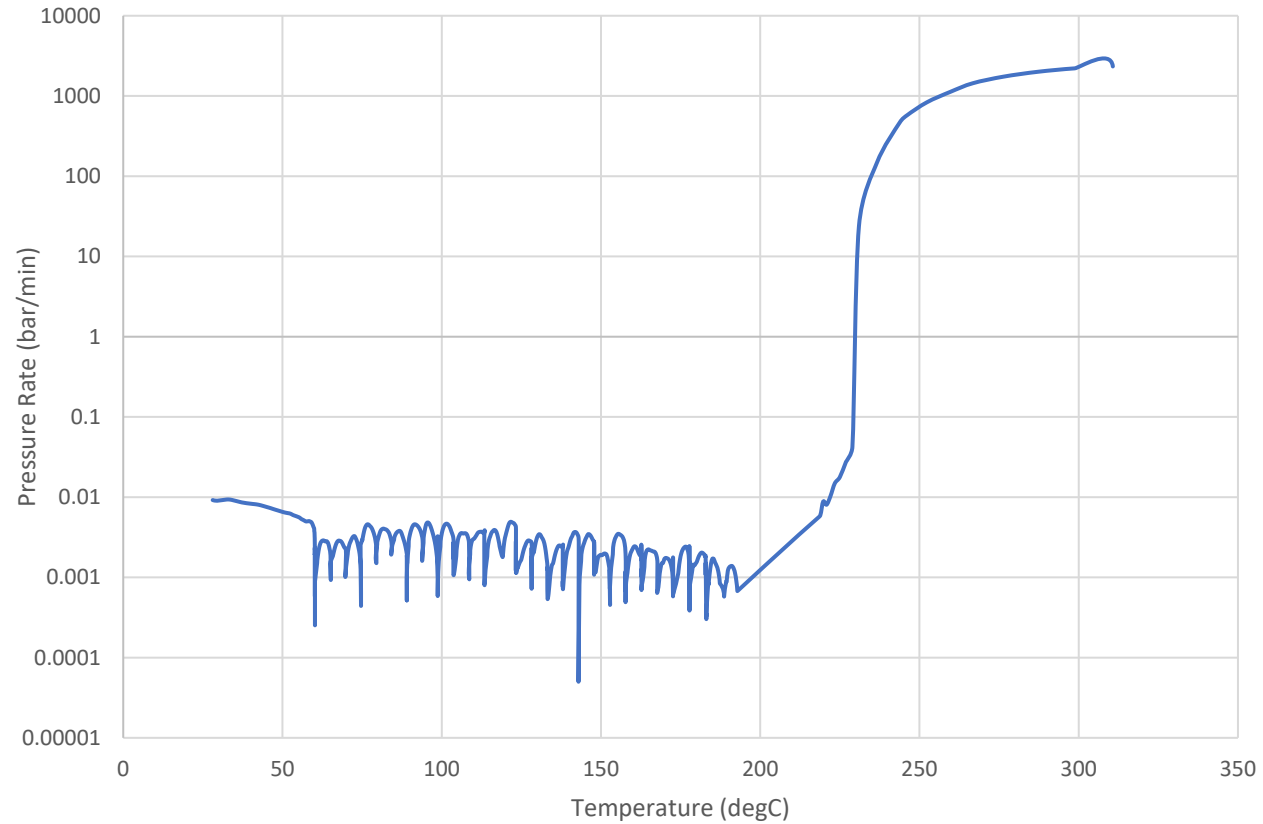


POSSIBLE SLIDES TO BE USED

Temperature Rate vs Temp



Pressure Rate vs Temp



SMALLER, LIGHTER SAMPLES ARE MORE ENERGETIC

