## **Spring 2018**

#### **ME-525: Combustion**

#### TTh 10:30 to 11:45 a.m.

### **Mechanical Engineering Building Room 1052**

**Instructor:** Jay P. Gore and Sameer V. Naik

Office: ME-1003H (Gore) and ME-G092 (Naik)

**Phone:** 765-494-0061 (Naik) and 765-496-2183 (Naik)

**Email:** gore@purdue.edu and naiks@purdue.edu

**Office Hours:** TTh 12 to 1:30 p.m. (Gore) and MW 8:30 to 10:00 a.m. (Naik)

If you cannot see the instructors during regular office hours, please email to arrange an appointment at another time. The use of email to ask questions is strongly encouraged. For email questions of general interest, the question and answer will be sent via email to the entire class after the name of the person who asked the original question has been removed.

TA: Vikrant Goyal
Office: ME-1030C

**Email:** goyal21@purdue.edu

**Office Hours:** MW 3 to 5 p.m.

**Textbook:** An Introduction to Combustion Concepts and Applications, Stephen R. Turns

3<sup>rd</sup> Edition, McGraw Hill, 2012

Grading: Homework 20%

Project 10% or 20%

Exam 1 20% Exam 2 20%

Final Exam 30% or 20%

**Academic dishonesty will not be tolerated.** Penalties include failing the course and will be reported to the university.

**Emergency:** Schedule and grading policy are subject to change in case of campus emergencies.

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# **Lectures and Reading Assignments**

**Textbook:** An Introduction to Combustion Concepts and Applications, Stephen R. Turns

Session	Date	Topics	Reading
1	Jan. 9	Introduction, Motivation, Energy, Fuels and	1-11, 638-679
		Alternate Fuels	
2	Jan. 11	Review of Property Relations, Ideal Gas Mixtures,	12-21
		1 <sup>st</sup> Law of Thermodynamics	
3	Jan. 16	Reactant and Product Mixtures, Enthalpy and	21-37
		Heating Values, Adiabatic Flame Temperature	
4	Jan. 18	Chemical Equilibrium, 2 <sup>nd</sup> Law of Thermodynamics,	38-66
		Products of Combustion	
5	Jan. 23	Chemical Kinetics, Global and Elementary	107-123
		Reactions, Rates and Types of Reactions	
6	Jan. 25	Chain Reactions, Time Scales, Catalysis,	123-140, Handout
		COSILAB Tutorial and Demonstration	
7	Jan. 30	Chemical Kinetic Mechanisms	149-170
8	Feb. 1	Coupling of Chemical and Thermal Analyses,	183-193
		Constant Volume Reactor, Constant Pressure Reactor	
9	Feb. 6	Well-Stirred Reactor and Plug Flow Reactor	194-211
10	Feb. 8	Mass and Species Conservation Equations	79-94, 220-226
11	Feb. 13	Binary and Multi-component Diffusion	226-233
12	Feb. 15	Momentum and Energy Conservation Equations	233-245
13	Feb. 20	Exam 1, Lectures 1 to 11, 6 to 8 p.m. in ME-1061	
14	Feb. 22	Conserved Scalars and Mixture Fraction	245-254
15	Feb. 27	Laminar Premixed Flame	258-276
16	Mar. 1	Laminar Premixed Flame Properties	276-287
17	Mar. 6	Quenching, Flammability, Ignition, Stabilization	287-303
18	Mar. 8	Detonations and Deflagrations: C-J Analysis	616-630
19	Mar. 20	Detonations and Deflagrations: ZND Structure	630-635

20	Mar. 22	Non-reacting Laminar Jet, Laminar Jet Diffusion	311-346	
		Flame Structure		
21	Mar. 27	Soot Formation, Counter-flow Flame Structure	346-359	
22	Mar. 29	Pollutants and Emissions	170-175, 556-602	
23	Apr. 3	Exam 2, Lectures 1 to 20, 6 to 8 p.m. in ME-1061		
24	Apr. 5	Introduction to Turbulent Flow	427-452	
25	Apr. 10	Turbulent Non-premixed Flames	486-526	
26	Apr. 12	Turbulent Premixed Flames	453-485	
27	Apr. 17	Carbon Particle Combustion	527-554	
28	Apr. 19	Droplet Evaporation and Burning	94-104, 366-382	
29	Apr. 24	Droplet Evaporation and Burning (contd.)	383-419	
30	Apr. 26	Spray Combustion		
Final Exam, Comprehensive, April 30 – May 5 (Date, Time, Location TBD)				