### Lecture #9

# **Environmentally Responsible**Design and Manufacturing

Prof. John W. Sutherland

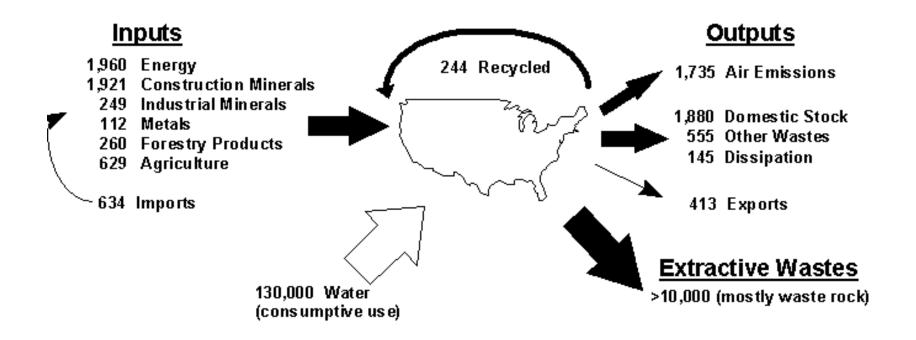
Jan. 30, 2004



# "National Material Metrics for Industrial Ecology," Wernick & Ausubel

all values in Mmt per year (1Mt = 1Gg)

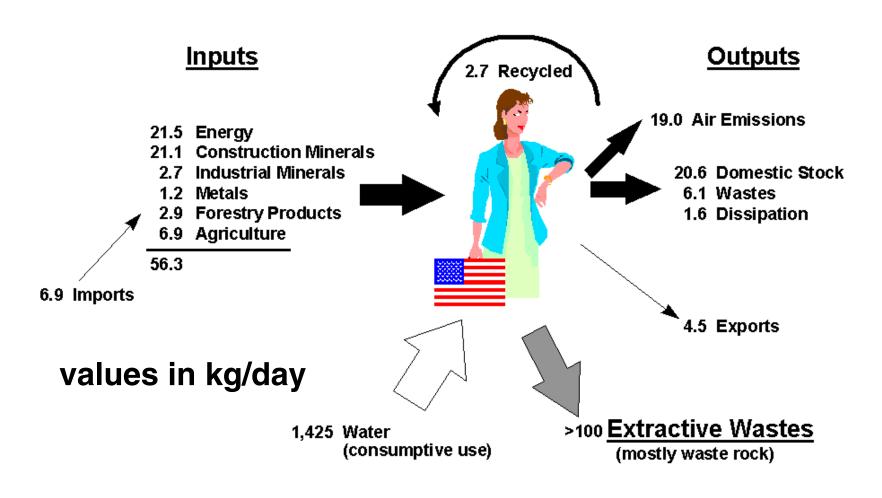
Figure 1





Environmentally Responsible Design & Manufacturing (MEEM 4685/5685) Dept. of Mechanical Engineering - Engineering Mechanics Michigan Technological University

Figure 2





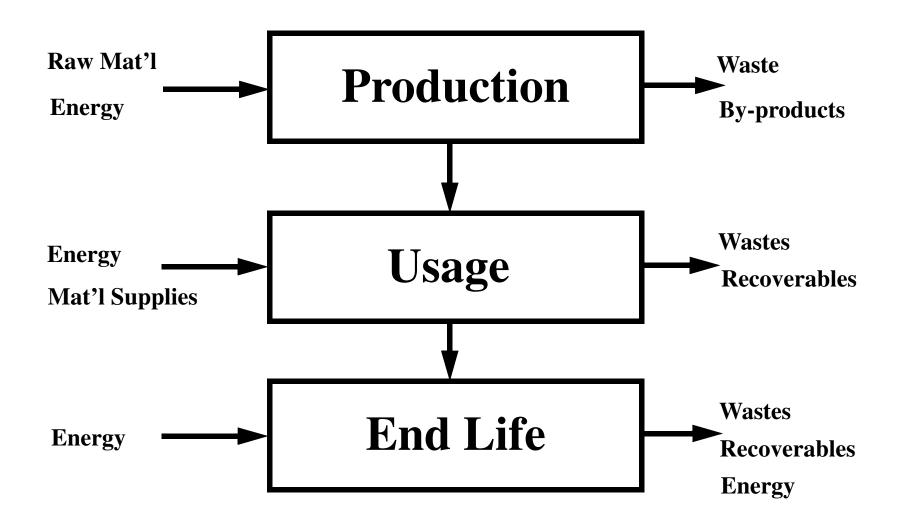
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# **Material Reserves**

Material	Range of Reserves at Current Price Levels	
Aluminum	3,247.7 - 4,726.4	Tg
Copper	335.7 - 459.9	Tg
Gold	31.1 - 57.5	Gg
Iron	87,724.8 - 186,880.1	Tg
Lead	87.1 - 130.6	Tg
Nickel	41.9 - 90.3	Tg
Tin	4.2 - 9.4	Tg
Titanium	143.3 - 296.6	Tg
Zinc	118.8 - 235.9	Tg



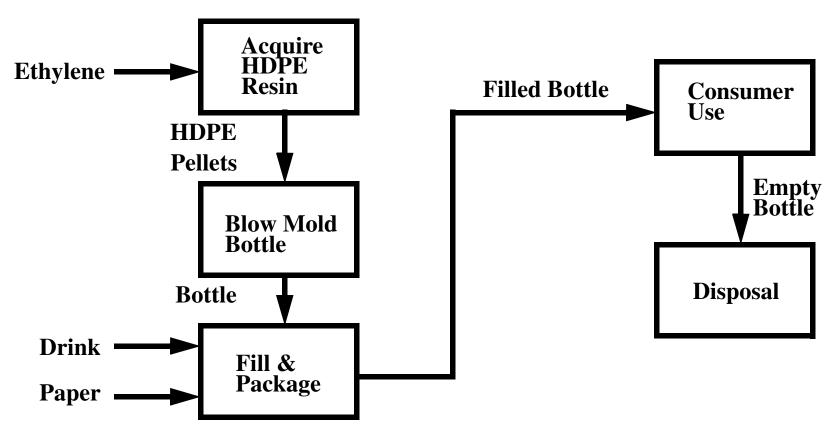
## **More on Flow Charts**





# Flow Charts (continued)

On the flowsheet indicate the inputs and the outputs on the arrows. Plastic Drink Bottle.





### More on Flow Charts

Once flowsheet is prepared, determine the data needed and where to get it (quality of the data?)

- number of bottles
- makeup of caps
- makeup of labels
- packing boxes
- foils
- post use of plastic
- energy usage



# More on Materials (courtesy of Prof. Rundman) \*\* also see the "materials.pdf" document \*\*

Material flow Diagram: Disposal in Landfill

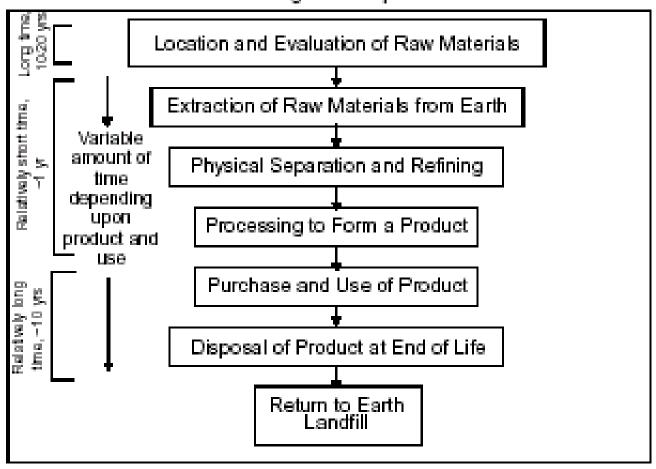




Figure 5-2 Material Flow Diagram: Recycle, Reuse, or Remanufacturing

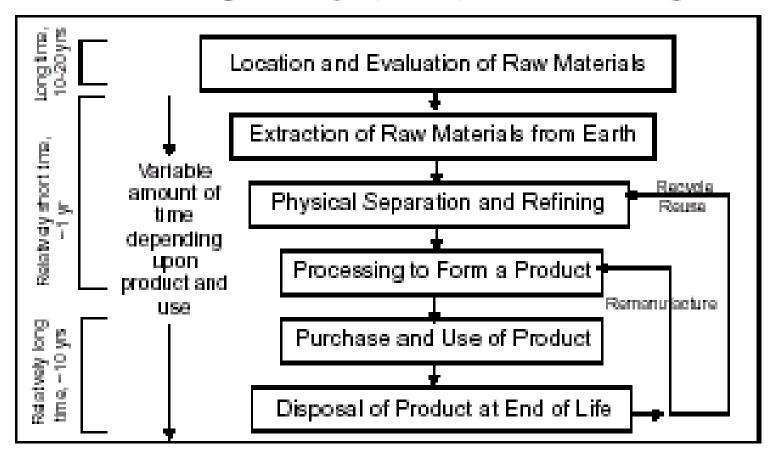




Table 7-2: World and U.S. Production Statistics of Primary Metals and Castings (in millions of metric tons)

Primary Metals*			Castings**		
Metal	World	U.S.	Metal	World	U.S.
Steel (97)	792.8	128.5	Cast Iron and Steel	59.6	11.7
Aluminum (96)	20.7	3.6	Aluminum base	6.37	1.64
Copper (97)	10.8	1.7	Copper base	1.03	0.31
Zinc (97)	7.7	0.45	Zinc base	0.81	0.37
Magnesium (96)	0.34	0.13	Magnesium base	0.05	0.03
Percent Ferrous	95.2	95.6		88	83

<sup>\*</sup>Data from *Metal Statistics* 1998, for the years 1996 & 1997, American Metal Market \*\*Data from *Modern Casting*, Dec. 1997, American Foundrymen's Society.



# **Steel Production**

- Mining
- Beneficiation -- Concentrating
- Pelletizing
- Reduction (in a Blast Furnace) -- remove oxygen
- Decarburizing (in Basic Oxygen Furnace) -- remove carbon



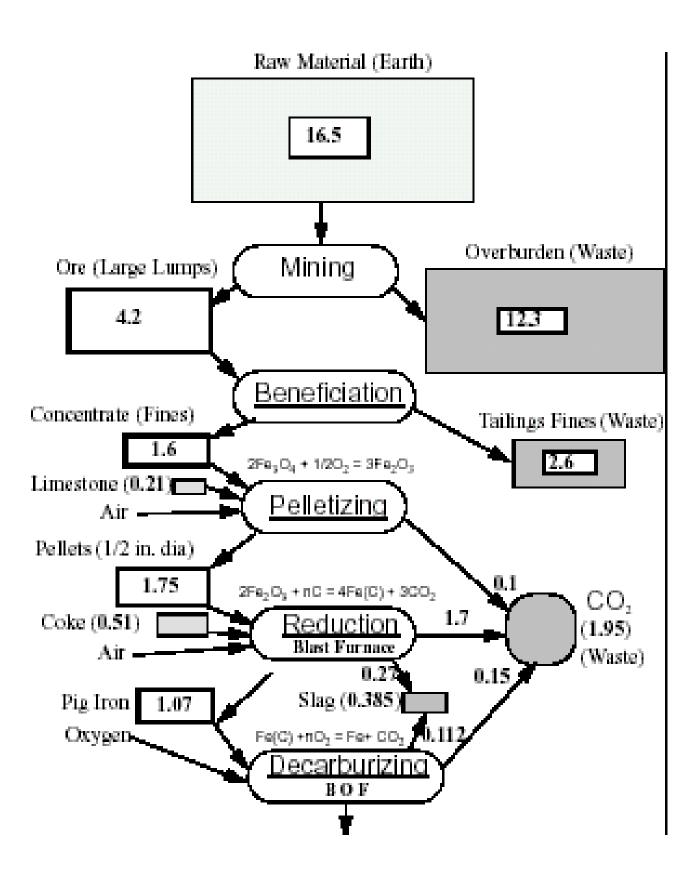
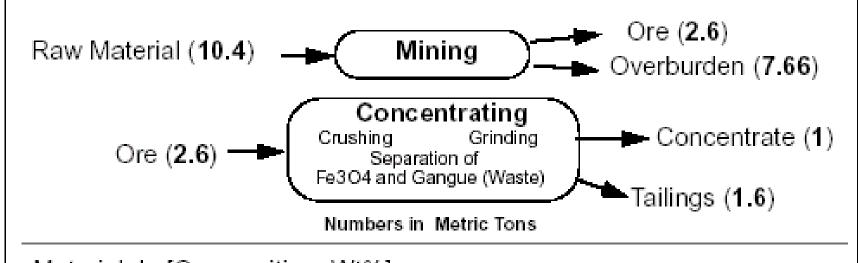


Figure 5-6 Simplified Mining and Beneficiation Mass Balances



Material In [Composition, Wt%]

Ore into concentrator [38.7 Fe<sub>3</sub>O<sub>4</sub>, 7.15 Fe<sub>2</sub>O<sub>3</sub>, 54.15 SiO<sub>2</sub>]

Material Out [Composition, Wt%]

Concentrated Ore [91.16 Fe<sub>3</sub>O<sub>4</sub>, 8.84 SiO<sub>2</sub>]

Tailings [6.0 Fe<sub>3</sub>O<sub>4</sub>, 11.6 Fe<sub>2</sub>O<sub>3</sub>, 82.4 SiO<sub>2</sub>]

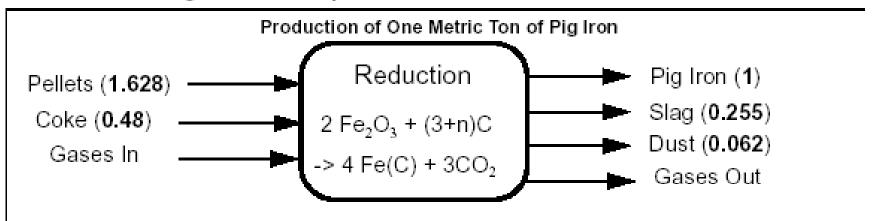


Requirements for Fluxed Pellets (Numbers in metric tons) Concentrate (0.92) Pelletizing CaCO<sub>3</sub> (0.089) Pellets (1) 2 Fe<sub>3</sub>O<sub>4</sub> + 1/2 O<sub>2</sub> ⇒ 3 Fe<sub>2</sub>O<sub>3</sub>  $MgCO_3(0.033)$ Gas (0.144) CaCO<sub>2</sub> -> CaO + CO<sub>2</sub> Bentonite (0.006) MgCO<sub>3</sub> -> MgO + CO<sub>2</sub> Air (0.096) Materials In [Composition, Wt %] Iron Ore Concentrate [91.2 Fe<sub>2</sub>O<sub>3</sub>, 8.8SiO<sub>2</sub>] Flux [72 CaCO<sub>3</sub>, 28 MgCO<sub>3</sub>] Air [23.3 O<sub>2</sub>, 76.7 N<sub>2</sub>] Bentonite [(OH)<sub>4</sub>Al<sub>4</sub>Si<sub>8</sub>O<sub>20</sub>] Materials Out [Composition, Wt %] Pellets [87.3 Fe<sub>2</sub>O<sub>2</sub>, 5.8 SiO<sub>2</sub>, 5CaO, 1.6 MgO, 0.3 Al<sub>2</sub>O<sub>3</sub>] Gas [CO<sub>2</sub>, N<sub>2</sub>]

Figure 5- 7 Simplified Pelletizing Materials Balance



Figure 5- 8 Simplified Blast Furnace Materials Balance



#### Materials In [Composition, Wt %]

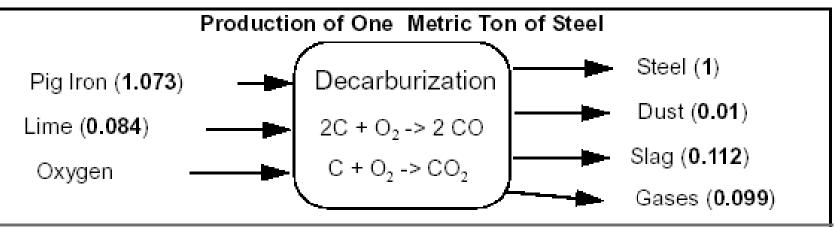
Pellets [  $87.3 \text{ Fe}_2\text{O}_3$ ,  $5.8 \text{ SiO}_2$ , 5 Cao, 1.6 MgO,  $0.3 \text{ Al}_2\text{O}_3$ ] Coke [90 C,  $6 \text{ SiO}_2$ ,  $3.5 \text{ Al}_2\text{O}_3$ , 0.5 CaO] Gases In [ $\text{O}_2$ ,  $\text{H}_2$ ,  $\text{CH}_4$ ,  $\text{C}_2\text{H}_6$ , CO,  $\text{CO}_2$ ]

#### Materials Out [Composition, Wt %]

Pig Iron [95 Fe, 4.2 C, 0.8 Si] Slag [ 48.3 SiO<sub>2</sub>, 33 CaO, 8.4 Al<sub>2</sub>O<sub>3</sub>, 10.3 MgO] Dust [Fe<sub>2</sub>O<sub>3</sub>] Gases Out [H<sub>2</sub>O, CO, CO<sub>2</sub>]



Figure 5- 9 Simplified (B O F) Materials Balance



#### Material In [Composition, Wt %]

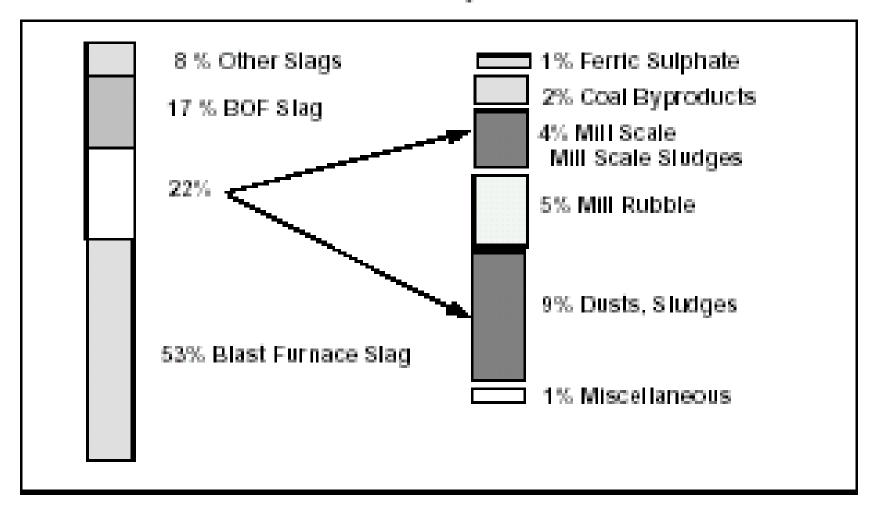
Pig Iron [95 Fe, 4.2 C, 0.8 Si] Lime [100CaO] Oxygen [100O<sub>2</sub>]

#### Material Out [Composition, Wt %]

Steel [99.3 Fe, 0.4 C, 0.3 Si] Slag [25 Fe<sub>2</sub>O<sub>3</sub>, 75 CaO] Dust [100Fe<sub>2</sub>O<sub>3</sub>] Gases [90 CO, 10 CO<sub>2</sub>]



#### Breakdown of the Waste Materials by Location in Steel Production



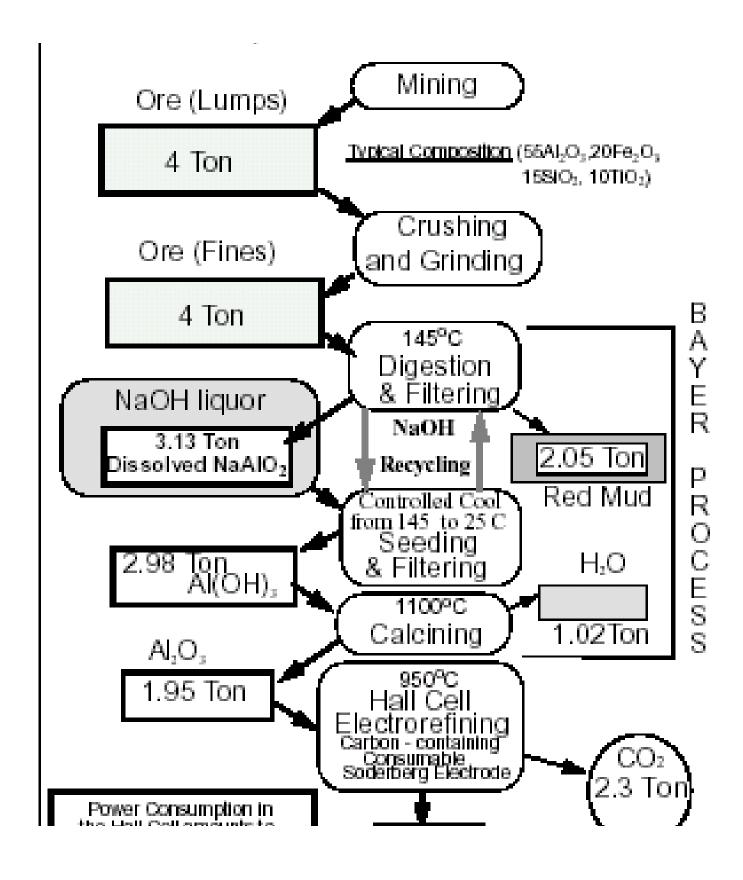


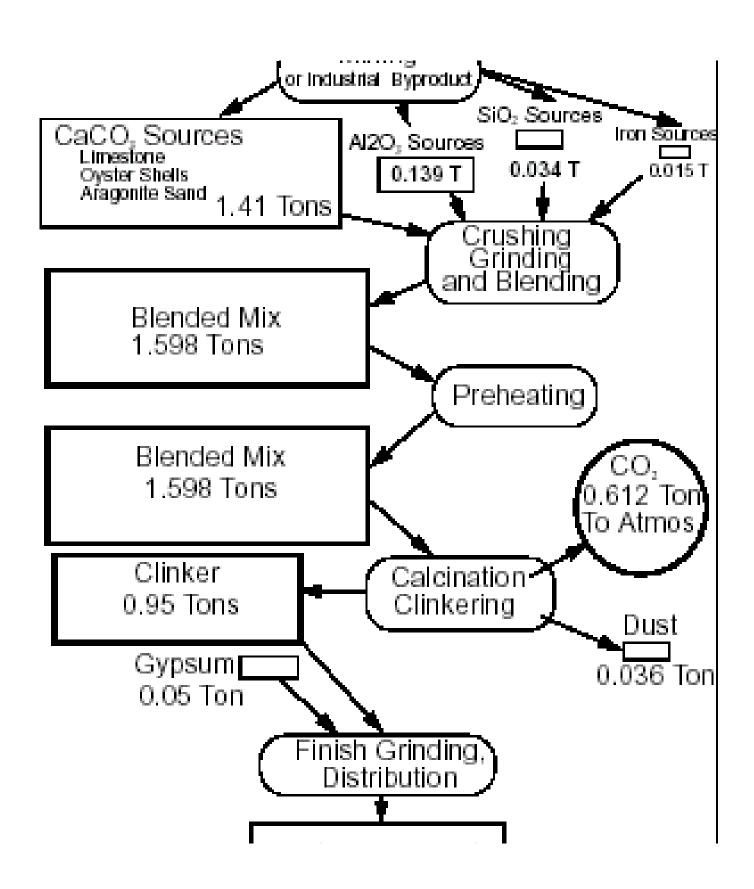
# **Copper Production**

- Mining Removal of Ore in Seams (1.2 wt Pct Cu) Leaving Waste Rock Underground
- 2. Beneficiation Production of Concentrate containing 30.5 wt. pct Copper. Discard Waste Rock in Tailings Pond
- 3. Smelting Production of Matte (62.5 wt pct Copper) in a coalfired reverberatory furnace. Discard Slag to Dump and SO2 Gas into the Atmosphere
- 4. Converting Production of Blister Copper (99 wt. pct Copper) with release of SO2 gas to Atmosphere and Slag Recycled to the Reverberatory Smelting Furnace
- 5. Fire Refining Deoxidizing the Blister Copper to Produce Anode Copper (99.7 wt. Pct Copper)
- 6. Anode Casting and Electro-refining Produce Anode Shape, Electro-refine to Produce 99.997 Wt. Pct Copper. Send Slimes to Recover Precious Metals (Ag, Au, Pt, etc.)



#### Numbers in Metholions Ore From Mining Operation 114 Tons (1.2 Wt. Pct. Cu) Beneficiation Tailings to Waste Pond 110.1 Tons (0.15 Wt Pct Cu) 3.9 Concentrate (30.5 Wt.Pct Cu) Smelting Slag (0.8 Wt Pct Cu) Matte 3.26 1.79 Converting (62 Wt Pct Cu) $SO_2$ Blister Cu 0.79 Tons Refining &AC (99 Wt. Pct Cu) Anode Cu 1.20 Electrorefine (99.7 Wt. Pct Cu)





# **Production of Plastics**

#### Feedstock Flowsheet

