PURDUE UNIVERSITY

Office of the Registrar FORM 40G REV. 7/08

REQUEST FOR ADDITION, EXPIRATION, OR REVISION OF A GRADUATE COURSE (50000-80000 LEVEL)

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FORM 40G REV. 7/08 OR REV

EPARTMENT School of Materials Engineering EFFECTIVE SESSION Spring 2040 2012 2013 INSTRUCTIONS: Please check the items below which describe the purpose of this request. New course with supporting documents (complete proposal form) 7. Change in course attributes 2. Add existing course offered at another campus 8. Change in instructional hours 3. Expiration of a course 9. Change in course description Change in course number 10. Change in course requisites Change in course title 5. 11. Change in semesters offered 6. Change in course credit/type 12. Transfer from one department to another PROPOSED: EXISTING: TERMS OFFERED Check All That Apply: Subject Abbreviation MSE Subject Abbreviation Summer Fall X Spring Course Number 507 56700 Course Number CAMPUS(ES) INVOLVED Calumet N. Central Long Title Polymer Synthesis Cont Ed Tech Statewide Ft. Wayne Short Title polymer synthesis XW. Lafayette Indianapolis Abbreviated title will be entered by the Office of the Registrar if omitted. (30 CHARACTERS ONLY) COURSE ATTRIBUTES: Check Ali That Apply 1.Fixed Credit: Cr. Hrs. 1. Pass/Not Pass Only 2.Variable Credit Range: 6. Registration Approval Type 2. Satisfactory/Unsatisfactory Only Department Minimum Cr. Hrs Instructor (Check One) 3. Repeatable Or 7. Variable Title Maximum Repeatable Credit Maximum Cr. Hrs. 8. Honors 4. Credit by Examination 9. Full Time Privilege 3.Equivalent Credit: No 5. Special Fees 4. Thesis Credit: Yes No 10. Off Campus Experience Schedule Type Minutes Weeks % of Credit Meetings Week 3 Allocated 100 Cross-Lis Courses Per Mtg 50 Offered ecture citation resentation Laboratory Lab Prep Studio Distance Clinic Experiential Research 25 Ind. Study Pract/Observ COURSE DESCRIPTION (INCLUDE REQUISITES/RESTRICTIONS): Sem. 2, Class 3, Cr. 3 (Offered in Alternate Years), Prerequisite: CHEM 257 Organic Chemistry Analyze properties of polymers and correlate them to chemical structure. Apply methods and mechanisms of polymer chemistry to industrial and laboratory synthesis. Examine the kinetic and design factors that control polymer structures. Professor Youngblood. Calumet Department Head Calumet School Dean Date Date Calumet Undergrad Curriculum Committee Date Fort Wayne Department Head Date Fort Wayne School Dean Fort Wayne Chancello Date Date Indianapolis Department Head Date Indianapolis School Dear Date Date. Date Graduate Dear Date



Supporting Document for a New Graduate Course

Purdue Universit	y Graduate (Council		(Select One)		
Faculty Member: Jeffrey P. Youngblood						
·		-		Reviewer:		
Campus:	West Lafaye	tte				
Required by the (Graduate Co			Comments:		
Contact for information if questions arise:		Name:	Jeffre	ey P. Youngblood		
		Phone Number: 4	496-2	496-2294		
				ungb@purdue.edu		
				MS 2233		
Course Subject Ab	breviation and	d Number: MSE 56	700			
Course Title: Poly	mer Synthesis					
	Faculty Member: Department: Campus: Proposal for New Required by the Campustrar's Form Contact for info questions arise: Course Subject Ab	Faculty Member: Jeffrey P. Yo Department: Materials En West Lafaye Proposal for New Graduate Co Required by the Graduate Co Registrar's Form 40G Contact for information if questions arise: Course Subject Abbreviation and	Department: Campus: Materials Engineering West Lafayette Proposal for New Graduate Course-Documentation Required by the Graduate Council to Accompany Registrar's Form 40G Contact for information if Name: questions arise: Phone Number: E-mail: Campus Address:	Faculty Member: Jeffrey P. Youngblood Department: Materials Engineering Campus: West Lafayette Proposal for New Graduate Course-Documentation Required by the Graduate Council to Accompany Registrar's Form 40G Contact for information if Name: Jeffre questions arise: Phone Number: 496-2 E-mail: jpyou Campus Address: ARM! Course Subject Abbreviation and Number: MSE 56700		

A. Justification for the Course:

- Provide a complete and detailed explanation of the need for the course (e. g., in the
 preparation of students, in providing new knowledge/training in one or more topics, in
 meeting degree requirements, etc.), how the course contributes to existing majors
 and/or concentrations, and how the course relates to other graduate courses offered by
 the department, other departments, or interdisciplinary programs.
- Justify the level of the proposed graduate course (50000- or 60000-level) including statements on, but not limited to: (1) the target audience, including the anticipated number of undergraduate and graduate students who will enroll in the course; and (2) the rigor of the course.

B. Learning Outcomes and Method of Evaluation or Assessment:

- Describe the course objectives and student learning outcomes that address the objectives (i.e., knowledge, communication, critical thinking, ethical research, etc.).
- Describe the methods of evaluation or assessment of student learning outcomes. (Include evidence for both direct and indirect methods.)
- Grading criteria (select from dropdown box); include a statement describing the criteria that will be used to assess students and how the final grade will be determined.

<u>Criteria</u>	Exams and Quizzes
	A come a come and a common consequence of a common and a

• Identify the method(s) of instruction (select from dropdown box) and describe how the methods promote the likely success of the desired student learning outcomes.

	T
Method of Instruction	Lecture

C. Prerequisite(s):

- List prerequisite courses by subject abbreviation, number, and title.
- List other prerequisites and/or experiences/background required. If no prerequisites are indicated, provide an explanation for their absence.

D. Course Instructor(s):

- Provide the name, rank, and department/program affiliation of the instructor(s).
- Is the instructor currently a member of the Graduate Faculty? Yes No (If the answer is no, indicate when it is expected that a request will be submitted.)

E. Course Outline:

• Provide an outline of topics to be covered and indicate the relative amount of time or emphasis devoted to each topic. If laboratory or field experiences are used to supplement a lecture course, explain the value of the experience(s) to enhance the quality of the course and student learning. For special topics courses, include a sample outline of a course that would be offered under the proposed course.

F. Reading List (including course text):

- A primary reading list or bibliography should be limited to material the students will be required to read in order to successfully complete the course. It should not be a compilation of general reference material.
- A secondary reading list or bibliography should include material students may use as background information.

G. Library Resources

- Describe the library resources that are currently available or the resources needed to support this proposed course.
- H. Example of a Course Syllabus (While not a necessary component of this supporting document, an example of a course syllabus is available, for information, by clicking on the link below, which goes to the Graduate School's Policies and Procedures Manual for Administering Graduate Student Programs. See Appendix K.)

http://www.gradschool.purdue.edu/downloads/Graduate School Policies and Procedures Manual.pdf

SUPPORTING DOCUMENT FOR MSE 56700 NEW GRADUATE COURSE

A. Justification for the Course:

The primary purpose of this course is to develop an understanding of how polymers are made synthetically, both in the laboratory and industrially, as well as understand critical factors such as reaction kinetics and polymer properties.

This class goes across a variety of disciplines (MSE, CHM, CHE & BME) and the content has been coordinated with MSE 525 (Structure-Property Relationships of Engineering Polymers) and CHE 544 (Structure and Physical Behavior of Polymer Systems). Expected enrollment would be between 10-15 graduate level students and 5-10 undergraduate level students.

The goals of this class are to analyze properties of a polymer and correlate them to chemical structure; to present methods and mechanisms of polymer chemistry and apply them to industrial and everyday realities; to examine the kinetic and design factors that control polymer structures.

B. Learning Outcomes and Method of Evaluation of Assessment:

Upon completion of the course, the student will understand, predict and design polymer properties based on polymer structure and how and why certain reactions take place through mechanisitic means. They will be able to produce a polymer structure based on known reactants and retrosynthetically design a chemical route to a variety of polymer structures. As well as, intuitively grasp the kinetic factors involved to predict such things a molecular weight, polydispersity, and others. They will understand polymer chemistry in relation to everyday materials and why industrial practice is as is.

Grading criteria will be derived from 3 mid-term exams and 1 comprehensive final and

- C. Prerequisites: CHEM 257, Organic chemistry
- D. Course Instructor: Professor Jeffrey P. Youngblood, Associate Professor of Materials Engineering and, currently, Chairman of the MSE Graduate Committee

E. Course Outline

<u>Week</u> <u>Topic</u>

- 1 Basic organic chemistry, basic polymer structure
- 2 T_g; T_m, gel point, geometric and stereo isomerism, MW, step vs. chain polymers
- 3 Step Polymers: esters, carbonates, amides, imides, urethanes
- 4 Step Polymers: phenolics, urea-formaldehyde, epoxy, silicones, specialties

Exam 1

- 5 Step Polymers: kinetics; Radical Chain: intro, T_c, initiation, propagation,
- 6 Radical Chain: termination, chain transfer, inhibition/retardation, acceleration,
- 7 Radical Chain: reaction kinetics; ATRP; TEMPO;
- 8 Copolymerization, Emulsion Polymerization

Exam 2

- 9 Anionic Chain: intro, initiation, termination, end control,
- Anionic Chain: structure; Cationic Chain: initiation, propagation, termination
- Anionic and cationic kinetics, ring opening polymerization
- 12 Basic organometallic chemistry

Exam 3

- 13 Catalysis: Ziegler-Natta;
- 14 Catalysis: olefin metathesis
- 15 Heteroatom Polymers, Conducting Polymers; Tie-in; review Final
- F. Reading List:

Textbook: "Principles of Polymerization", 4th Ed., G. Odian (Wiley-Itnerscience, 2004)

- G. Library Resources: No library resources needed
- H. Example of Course Syllabus

Engineering Faculty Document No. 31-09 December 3, 2008

TO:

The Engineering Faculty

FROM:

The Faculty of the School Materials Engineering

RE:

New Dual-Level Course, MSE 56700

The faculty of the School of Materials Engineering have approved the following new course and is being submitted to the Engineering Faculty with a recommendation for approval:

MSE 56700

Polymer Synthesis

Sem. 2. Class 3, Cr. 3. (Offered in Alternate Years) Prerequisites: CHEM 25700 Organic Chemistry

Description:

Analyze properties of polymers and correlate them to chemical structure. Apply methods and mechanisms of polymer chemistry to industrial and laboratory synthesis. Examine the kinetic and design factors that control polymer structures.

Reason: This class has now been offered three times as MSE 597Y, in Fall 2004 (25 students), Fall 2006 (17 students), and Spring 2008 (16 students) with good enrollment from students across a variety of disciplines (MSE, Chem, ChemE, BME). The primary purpose of this course is to develop an understanding of how polymers are made synthetically, both in the laboratory and industrially, as well as understand critical factors such as reaction kinetics and polymer properties. The techniques cover a large design space of polymers, however, the focus is on commodity, engineering and specialty resins to give an insight into the importance of the respective material that is being discussed. The content of this course has been coordinated with MSE 52500 (Structure-Property Relationships of Engineering Polymers) and CHE 54400 (Structure and Physical Behavior of Polymer Systems).

Sincerely yours,

Keith J. Bowman Head and Professor School of Materials Engineering Professor, School of Engineering Education (by courtesy) ABBROVER EAB THE EAGULTY
OF THE SCHOOLS OF ENGINEERING
BY THE ENGINEERING
CURRICULUM COMMITTEE

ECC Minutes #23

~ KFA ------

Chairman ECC

Supporting Documentation Page 1 or 2

MSE 597Y

Polymer Synthesis

Instructors: Jeffrey P. Youngblood, jpyoungb@purdue.edu, ARMS 2233, 496-2294

Course Description: Chemical synthetic routes to preparation of various polymers as well as the kinetic expressions for their reactions. Relations to industrial versus laboratory preparation are explored. Control of structure (chemical, tacticity, etc) and how this applies to properties is also taught.

Prerequisite: CHEM 257 Organic Chemistry

Goals:

To analyze properties of a polymer and correlate them to chemical structure. To present methods and mechanisms of polymer chemistry and apply them to industrial and everyday realities. To examine the kinetic and design factors that control polymer structures

Objectives:

Upon completion of the course, the student is able to:

- Understand, predict and design polymer properties based on polymer structure
- Understand how and why certain reactions take place through mechanistic means.
- Produce a polymer structure based on known reactants and retrosynthetically design a chemical route to a variety of polymer structures
- Intuitively grasp the kinetic factors involved to predict such things as molecular weight, polydispersity, and others.
- Understand polymer chemistry in relation to everyday materials and why industrial practice is as is.

Textbook:

"Principles of Polymerization", 4th Ed., G. Odian (Wiley-Interscience, 2004).

Professional category content as estimated by faculty member who prepared this course description:

Engineering Science: 3 Credits

Assessment:

3 mid-term exams (20% each) and 1 comprehensive final (40%)

Supporting Documentation Page 2 of 2

Spring 2008 Syllabus

<u>Week</u>	<u>Topic</u>
1	Basic organic chemistry, basic polymer structure
2	T _g ; T _m , gel point, geometric and stereo isomerism, MW, step vs. chain polymers
3	Step Polymers: esters, carbonates, amides, imides, urethanes
4	Step Polymers: phenolics, urea-formaldehyde, epoxy, silicones, specialties
	Exam 1
5	Step Polymers: kinetics; Radical Chain: intro, T _c , initiation, propagation,
6	Radical Chain: termination, chain transfer, inhibition/retardation, acceleration,
7	Radical Chain: reaction kinetics; ATRP; TEMPO;
8	Copolymerization, Emulsion Polymerization
	Exam 2
9	Anionic Chain: intro, initiation, termination, end control,
10	Anionic Chain: structure; Cationic Chain: initiation, propagation, termination
11	Anionic and cationic kinetics, ring opening polymerization
12	Basic organometallic chemistry
	Exam 3
13	Catalysis: Ziegler-Natta;
14	Catalysis: olefin metathesis
15	Heteroatom Polymers, Conducting Polymers; Tie-in; review
	Final

Engineering Faculty Document No. 31-09 December 3, 2008

TO: The Engineering Faculty

FROM: The Faculty of the School Materials Engineering

RE: New Dual-Level Course, MSE 56700

The faculty of the School of Materials Engineering have approved the following new course and is being submitted to the Engineering Faculty with a recommendation for approval:

MSE 56700 Polymer Synthesis

Sem. 2. Class 3, Cr. 3. (Offered in Alternate Years) Prerequisites: CHEM 25700 Organic Chemistry

Description: Analyze properties of polymers and correlate them to chemical structure. Apply

methods and mechanisms of polymer chemistry to industrial and laboratory synthesis. Examine the kinetic and design factors that control polymer structures.

Reason: This class has now been offered three times as MSE 597Y, in Fall 2004 (25 students), Fall 2006 (17 students), and Spring 2008 (16 students) with good enrollment from students across a variety of disciplines (MSE, Chem, ChemE, BME). The primary purpose of this course is to develop an understanding of how polymers are made synthetically, both in the laboratory and industrially, as well as understand critical factors such as reaction kinetics and polymer properties. The techniques cover a large design space of polymers, however, the focus is on commodity, engineering and specialty resins to give an insight into the importance of the respective material that is being discussed. The content of this course has been coordinated with MSE 52500 (Structure-Property Relationships of Engineering Polymers) and CHE 54400 (Structure and Physical Behavior of Polymer Systems).

Sincerely yours,

Keith J. Bowman Head and Professor School of Materials Engineering Professor, School of Engineering Education (by courtesy) ABBRAVER EAB THE FACULTY
OF THE SCHOOLS OF ENGINEERING
BY THE ENGINEERING
CURRICULUM COMMITTEE

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Chairman ECC

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MSE 597Y

Polymer Synthesis

Instructors: Jeffrey P. Youngblood, jpyoungb@purdue.edu, ARMS 2233, 496-2294

Course Description: Chemical synthetic routes to preparation of various polymers as well as the kinetic expressions for their reactions. Relations to industrial versus laboratory preparation are explored. Control of structure (chemical, tacticity, etc) and how this applies to properties is also taught.

Prerequisite: CHEM 257 Organic Chemistry

Goals:

To analyze properties of a polymer and correlate them to chemical structure. To present methods and mechanisms of polymer chemistry and apply them to industrial and everyday realities. To examine the kinetic and design factors that control polymer structures

Objectives:

Upon completion of the course, the student is able to:

- Understand, predict and design polymer properties based on polymer structure
- Understand how and why certain reactions take place through mechanistic means.
- Produce a polymer structure based on known reactants and retrosynthetically design a chemical route to a variety of polymer structures
- Intuitively grasp the kinetic factors involved to predict such things as molecular weight, polydispersity, and others.
- Understand polymer chemistry in relation to everyday materials and why industrial practice is as is.

Textbook:

"Principles of Polymerization", 4th Ed., G. Odian (Wiley-Interscience, 2004).

Professional category content as estimated by faculty member who prepared this course description:

Engineering Science: 3 Credits

Assessment:

3 mid-term exams (20% each) and 1 comprehensive final (40%)

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Spring 2008 Syllabus

<u>Week</u>	<u>Topic</u>
1	Basic organic chemistry, basic polymer structure
2	T _g ; T _m , gel point, geometric and stereo isomerism, MW, step vs. chain polymers
3	Step Polymers: esters, carbonates, amides, imides, urethanes
4	Step Polymers: phenolics, urea-formaldehyde, epoxy, silicones, specialties
	Exam 1
5	Step Polymers: kinetics; Radical Chain: intro, T _c , initiation, propagation,
6	Radical Chain: termination, chain transfer, inhibition/retardation, acceleration,
7	Radical Chain: reaction kinetics; ATRP; TEMPO;
8	Copolymerization, Emulsion Polymerization
	Exam 2
9	Anionic Chain: intro, initiation, termination, end control,
10	Anionic Chain: structure; Cationic Chain: initiation, propagation, termination
11	Anionic and cationic kinetics, ring opening polymerization
12	Basic organometallic chemistry
	Exam 3
13	Catalysis: Ziegler-Natta;
14	Catalysis: olefin metathesis
15	Heteroatom Polymers, Conducting Polymers; Tie-in; review
	Final

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Office of the Registrar FORM 40G REV. 7/08

Clinic

PURDUE UNIVERSITY

REQUEST FOR ADDITION, EXPIRATION. OR REVISION OF A GRADUATE COURSE (50000-80000 LEVEL)

Print Form

EFD 31-09

DEPARTMENT School of Materials Engineering **EFFECTIVE SESSION** Spring 2010 INSTRUCTIONS: Please check the Items below which describe the purpose of this request. New course with supporting documents (complete proposal form) 7. Change in course attributes 2. Add existing course offered at another campus 8. Change in instructional hours Expiration of a course 9. Change in course description 10. Change in course requisites Change in course number Change in course title 11. Change in semesters offered 6. Change in course credit/type 12. Transfer from one department to another PROPOSED: **EXISTING:** TERMS OFFERED Check All That Apply: Subject Abbreviation MSE Subject Abbreviation 7 Summer Fall **Spring** Course Number 567 Course Number CAMPUS(ES) INVOLVED Calumet N. Central Long Title Polymer Synthesis Cont Ed Tech Statewide Ft. Wayne Short Title polymer synthesis XW. Lafayette Indianapolis Abbreviated title will be entered by the Office of the Registrar if omitted. (30 CHARACTERS ONLY) CREDIT TYPE COURSE ATTRIBUTES: Check All That Apply 1.Fixed Credit: Cr. Hrs. 1. Pass/Not Pass Only 6. Registration Approval Type 2. Variable Credit Range: 2. Satisfactory/Unsatisfactory Only Minimum Cr. Hrs Department Instructor (Check One) 3. Repeatable Or 7. Variable Title Maximum Repeatable Credit: Maximum Cr. Hrs. 8. Honors 4. Credit by Examination 3.Equivalent Credit: Yes 9. Full Time Privilege No 5. Special Fees 4. Thesis Credit: No 10. Off Campus Experience Schedule Type Weeks % of Credit Meetings Per Offered 16 Cross-Listed Courses Aliocated 100 Lecture ecitation resentation Laboratory Lab Prep Studio Distance Experiential Research Ind. Study Pract/Observ COURSE DESCRIPTION (INCLUDE REQUISITES/RESTRICTIONS): Sem. 2, Class 3, Cr. 3 (Offered in Alternate Years), Prerequisite: CHEM 257 Organic Chemistry

Analyze properties of polymers and correlate them to chemical structure. Apply methods and mechanisms of polymer chemistry to industrial and laboratory synthesis. Examine the kinetic and design factors that control polymer structures.

Calumet Department Head Calumet School Dean Data Calumet Undergrad Curriculum Committee Date Date Fort Wayne Department Head Fort Wayne Chancello Fort Wayne School Dean Date Date Date 2010 Indianapotis Dipartment Head Indianapolis School Dear Date Date Date Date Date Approved by Graduate Council Graduate Council Secretary Date iduate Area Committee Convener Graduate Dean West Lafayette Registrar Date Date Date

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Supporting Document for a New Graduate Course

To:	Purdue University Graduate Council			(Select One)		
From:	Faculty Member: Jeffrey P. Youngblood					
	Department:	Materials Er	gineering		D	
	Campus:	West Lafaye	tte		Reviewer:	
Date:						
Subject:		Graduate Co	Course-Documentati uncil to Accompan		Comments:	
	Contact for information if questions arise:		Name:	Jeffre	ey P. Youngblood	
			E-mail: jpyo		496-2294	
					ungb@purdue.edu	
		S 2233				
	Course Subject Ab	breviation an	d Number: MSE 56	700		
	Course Title: Poly	mer Synthesis	3			

A. Justification for the Course:

- Provide a complete and detailed explanation of the need for the course (e. g., in the preparation of students, in providing new knowledge/training in one or more topics, in meeting degree requirements, etc.), how the course contributes to existing majors and/or concentrations, and how the course relates to other graduate courses offered by the department, other departments, or interdisciplinary programs.
- Justify the level of the proposed graduate course (50000- or 60000-level) including statements on, but not limited to: (1) the target audience, including the anticipated number of undergraduate and graduate students who will enroll in the course; and (2) the rigor of the course.

B. Learning Outcomes and Method of Evaluation or Assessment:

- Describe the course objectives and student learning outcomes that address the objectives (i.e., knowledge, communication, critical thinking, ethical research, etc.).
- Describe the methods of evaluation or assessment of student learning outcomes. (Include evidence for both direct and indirect methods.)
- Grading criteria (select from dropdown box); include a statement describing the criteria that will be used to assess students and how the final grade will be determined.

Criteria	Exams and Quizzes
Officeria	LAUTIS UTIC QUIZZES

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•	Identify the method(s) of instruction (select from dropdown box) and describe how the
	methods promote the likely success of the desired student learning outcomes.

Method of Instruction	Lecture
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C. Prerequisite(s):

- List prerequisite courses by subject abbreviation, number, and title.
- List other prerequisites and/or experiences/background required. If no prerequisites are indicated, provide an explanation for their absence.

D. Course Instructor(s):

- Provide the name, rank, and department/program affiliation of the instructor(s).
- Is the instructor currently a member of the Graduate Faculty? ___ Yes ___ No (If the answer is no, indicate when it is expected that a request will be submitted.)

E. Course Outline:

• Provide an outline of topics to be covered and indicate the relative amount of time or emphasis devoted to each topic. If laboratory or field experiences are used to supplement a lecture course, explain the value of the experience(s) to enhance the quality of the course and student learning. For special topics courses, include a sample outline of a course that would be offered under the proposed course.

F. Reading List (including course text):

- A primary reading list or bibliography should be limited to material the students will be required to read in order to successfully complete the course. It should not be a compilation of general reference material.
- A secondary reading list or bibliography should include material students may use as background information.

G. Library Resources

- Describe the library resources that are currently available or the resources needed to support this proposed course.
- H. Example of a Course Syllabus (While not a necessary component of this supporting document, an example of a course syllabus is available, for information, by clicking on the link below, which goes to the Graduate School's Policies and Procedures Manual for Administering Graduate Student Programs. See Appendix K.)

http://www.gradschool.purdue.edu/downloads/Graduate School Policies and Procedures Manual.pdf

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SUPPORTING DOCUMENT FOR MSE 56700 NEW GRADUATE COURSE

A. Justification for the Course:

The primary purpose of this course is to develop an understanding of how polymers are made synthetically, both in the laboratory and industrially, as well as understand critical factors such as reaction kinetics and polymer properties.

This class goes across a variety of disciplines (MSE, CHM, CHE & BME) and the content has been coordinated with MSE 525 (Structure-Property Relationships of Engineering Polymers) and CHE 544 (Structure and Physical Behavior of Polymer Systems). Expected enrollment would be between 10-15 graduate level students and 5-10 undergraduate level students.

The goals of this class are to analyze properties of a polymer and correlate them to chemical structure; to present methods and mechanisms of polymer chemistry and apply them to industrial and everyday realities; to examine the kinetic and design factors that control polymer structures.

B. Learning Outcomes and Method of Evaluation of Assessment:

Upon completion of the course, the student will understand, predict and design polymer properties based on polymer structure and how and why certain reactions take place through mechanisitic means. They will be able to produce a polymer structure based on known reactants and retrosynthetically design a chemical route to a variety of polymer structures. As well as, intuitively grasp the kinetic factors involved to predict such things a molecular weight, polydispersity, and others. They will understand polymer chemistry in relation to everyday materials and why industrial practice is as is.

Grading criteria will be derived from 3 mid-term exams and 1 comprehensive final and

- C. Prerequisites: CHEM 257, Organic chemistry
- D. Course Instructor: Professor Jeffrey P. Youngblood, Associate Professor of Materials Engineering and, currently, Chairman of the MSE Graduate Committee

E. Course Outline

Week Topic Basic organic chemistry, basic polymer structure T_g; T_m, gel point, geometric and stereo isomerism, MW, step vs. chain polymers Step Polymers: esters, carbonates, amides, imides, urethanes Step Polymers: phenolics, urea-formaldehyde, epoxy, silicones, specialties

Exam 1

- 5 Step Polymers: kinetics; Radical Chain: intro, T_c, initiation, propagation,
- 6 Radical Chain: termination, chain transfer, inhibition/retardation, acceleration,
- 7 Radical Chain: reaction kinetics; ATRP; TEMPO;
- 8 Copolymerization, Emulsion Polymerization

Exam 2

- 9 Anionic Chain: intro, initiation, termination, end control,
- 10 Anionic Chain: structure; Cationic Chain: initiation, propagation, termination
- Anionic and cationic kinetics, ring opening polymerization
- 12 Basic organometallic chemistry

Exam 3

- 13 Catalysis: Ziegler-Natta;
- 14 Catalysis: olefin metathesis
- 15 Heteroatom Polymers, Conducting Polymers; Tie-in; review

Final

F. Reading List:

Textbook: "Principles of Polymerization", 4th Ed., G. Odian (Wiley-Itnerscience, 2004)

- G. Library Resources
- H. Example of Course Syllabus

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Office of the Registrar FORM 40G REV. 7/08

PURDUE UNIVERSITY

REQUEST FOR ADDITION, EXPIRATION, OR REVISION OF A GRADUATE COURSE (50000-80000 LEVEL)

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EFD 31-09

DEPARTMENT School of Materials Engineering	E	FFECTIVE SESSION Spring	2010	,
INSTRUCTIONS: Please check the items below	which describe the purpose of this	request.		
1. New course with supporting docu 2. Add existing course offered at an 3. Expiration of a course 4. Change in course number 5. Change in course title 6. Change in course credit/type		7. 8. 9. 10. 11.	Change in course attributional in Change in course described change in course requisionage in course requisionage in semesters off Transfer from one depart	nours iption sites fered
PROPOSED:	EXISTING:			RMS OFFERED
Subject Abbreviation MSE	Subject Abbreviation		Che	ck All That Apply:
Course Number 587	Course Number		<u> </u>	US(ES) INVOLVED
Long Title Polymer Synthesis			Calumet Cont Ed	N. Central
Short Title polymer synthesis	· · · · · · · · · · · · · · · · · · ·	······································	Ft. Wayne	Tech Statewide
Abbreviated title will be entered	by the Office of the Registrar if on	nitted. (30 CHARACTERS ONLY)	Indianapolis	
CREDIT TYPE		COURSE ATTRIBUTES: 0	heck All That Apply	
1.Fixed Credit: Cr. Hrs. 3 2.Variable Credit Range: Minimum Cr. Hrs (Check One) To	Offered Allocated 100 S/RESTRICTIONS):), Prerequisite: CHEM 257 Organin to chemical structure. Apply metals.	7. Vari	Time Privilege Campus Experience	Cross-Listed Courses
Calumet Department Head Date	Calumet School Dean	Date	Calumet Undergrad Curricu	tum Committee Date
Fort Wayne Department Head Date	Fort Wayne School Dean	Date	Fort Wayne Chancelor	2/ Date
Indianapolis Department Head Date	Indianapolis School Dean	Date	Undergrad Cumiculum Com	2/16/2010 Date Date
North Control Department Head Date	North Central Chancellor	Date, 2/10	Date Approved by Graduate	• Council
West Variayette Department Head Bate	West Lafayette College/School Dean		Graduate Council Secretary	
raduate Area Committee Convener Date	Graduate Dean	Date	West Latayette Registrar	Date

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SCHOOL OF MATERIALS ENGINEERING

Engineering Faculty Document No. 31-09 December 3, 2009

TO:

The Engineering Faculty

FROM:

The Faculty of the School Materials Engineering

DATE:

November 17, 2008

RE:

New Dual-Level Course, MSE 567

The faculty of the School of Materials Engineering have approved the following new course and is being submitted to the Engineering Faculty with a recommendation for approval:

MSE 567

Polymer Synthesis

Sem. 2. Class 3, Cr. 3. (Offered in Alternate Years) Prerequisites: CHEM 257 Organic Chemistry

Description:

Analyze properties of polymers and correlate them to chemical structure. Apply methods and mechanisms of polymer chemistry to industrial and laboratory synthesis. Examine

the kinetic and design factors that control polymer structures.

Reason: This class has now been offered three times, in Fall 2004 (25 students), Fall 2006 (17 students), and Spring 2008 (16 students) with good enrollment from students across a variety of disciplines (MSE, Chem, ChemE, BME). The primary purpose of this course is to develop an understanding of how polymers are made synthetically, both in the laboratory and industrially, as well as understand critical factors such as reaction kinetics and polymer properties. The techniques cover a large design space of polymers, however, the focus is on commodity, engineering and specialty resins to give an insight into the importance of the respective material that is being discussed. The content of this course has been coordinated with MSE 525 (Structure-Property Relationships of Engineering Polymers) and CHE 544 (Structure and Physical Behavior of Polymer Systems).

Sincerely yours,

Keith J. Bowman Head and Professor

School of Materials Engineering

Professor, School of Engineering Education

(by courtesy)

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