

*Structural Changes in the  
Selectivity Filter of the Open-  
State KcsA channel*

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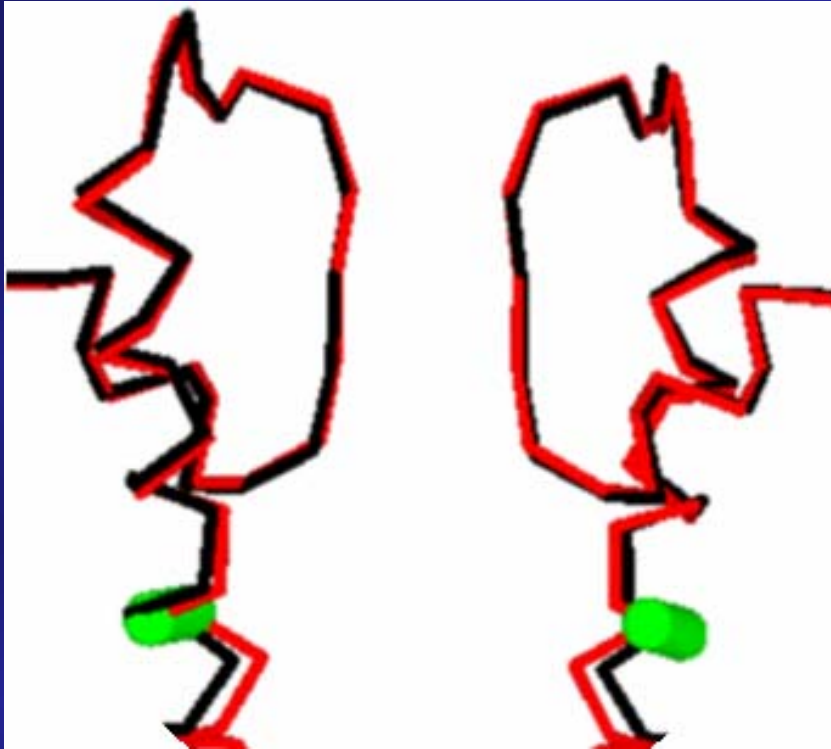
# Outline

1. **Key Concepts & Model**
2. **Conformations of the SF**
3. **Ionic Selectivity in KcsA**

*Telluride, CO*

*July 31, 2007*

# *Conformation of the SF P-loops*



Superposed C $_{\alpha}$  traces of the SF P-loops. **Closed** structure shown in **red**, open structure shown in **black**. **Green marks** show the location of a gating hinge, G99.

- SF P-loops are structurally unaltered in the open-state KcsA channel

*Therefore*

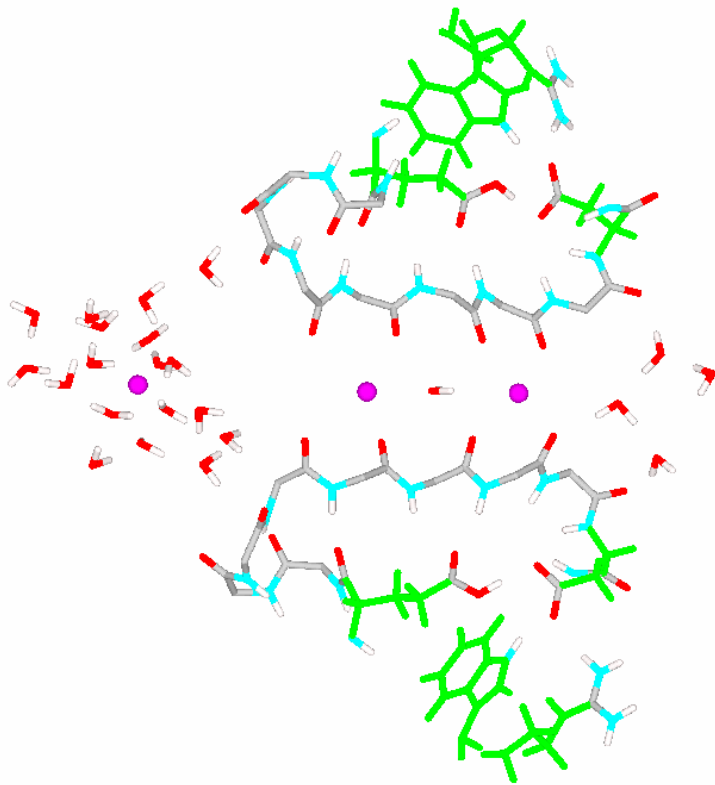
- closed-state crystal structure (1K4C) of the SF can be used as a reference state :

to determine structural changes in response to mutations or to  $K^{+} \Rightarrow Na^{+}$  substitution in the SF of the open-state KcsA

# *Conformation of the SF P-loops*

- **primary gate** – the IC bundle; **secondary gate** – the selectivity filter
- are the two gates independent or they somehow coupled?

Cordero-Morales et al. *Nat. Struct. Mol. Biol.* **13**, 311-318 (2006);  
Blunck et al. *J. Gen. Physiol.* **128**, 569-581 (2006)



The movie shows conformations of the SF P-loops and residues during the closed-open gating transition

- SF P-loops and residues fluctuate near the initial conformation
- side chains of R89 and W67 undergo deviations during gating, but they stay close to the initial conformation in the open state
- **no crucial structural changes are observed that could inactivate the SF**

***Ultimate goal:*** *use a computational approach to predict:*

*mutant protein structures and local structural changes due to substitutions, insertions or deletions of residues, waters and ions in the SF of the open-state KcsA*

### ***MD simulations?***

- MD yields trajectories that provide a comprehensive description of molecular motion and appear quite complicated resembling "random noise"
- trajectories do not go to a final well-tuned conformation, but represent the fluctuating protein with small-scale conformational transitions
- thermal fluctuations cause the protein structure to wander away from the crystal structure making comparisons difficult
- **MD simulations are not well suitable for our goal**

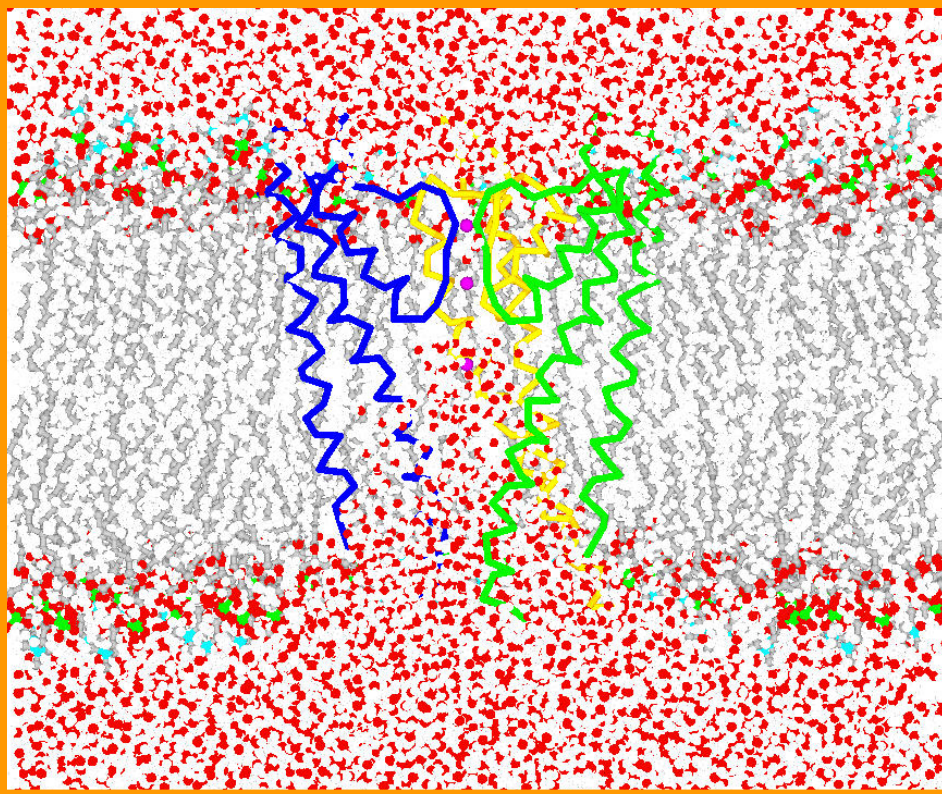


## *Why should energy minimization work?*

*Because ...*

- a protein's native-fold undergoes only *local* structural adjustments at mutation sites in the SF during minimization, and deviates somewhat *locally* from the experimental structure; no major repacking of the rest of the KcsA protein occurs
- the final conformation of the energy-minimized mutant is also a low-temperature structure, similar to the X-ray crystallographic structure; thus, structural refinement of the mutant is performed relative to low-temperature data with *local* adjustments at mutation sites
- energy minimization fine tunes (refines) the location, the optimal geometry and the energy of the mutant protein's preferred conformation that can be used for comparisons with the available experimental structures

# *Computational Model*



Side view of the central simulation cell showing open-state KcsA, 3 ions, 123 DPPC lipids and ~7,000 waters (~400 in the IC vestibule). Front lipids and one KcsA subunit are not displayed.

- the open-state KcsA structure with the cavity as an integral part of the cytoplasm is used

- periodic boundary conditions applied in all dimensions; the Coulomb energy calculated by a parallelized FMM; minimizations were performed using the steepest descent method and a new conjugate gradient method with guaranteed descent

- the large solvent box:
  - 1) stabilizes polar and charged residues on the surface of KcsA;
  - 2) accounts for surroundings' effect on SF conformations

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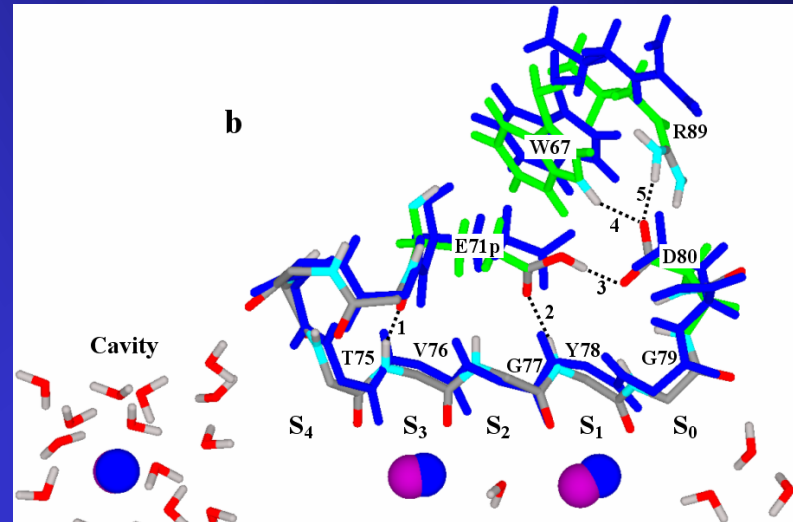
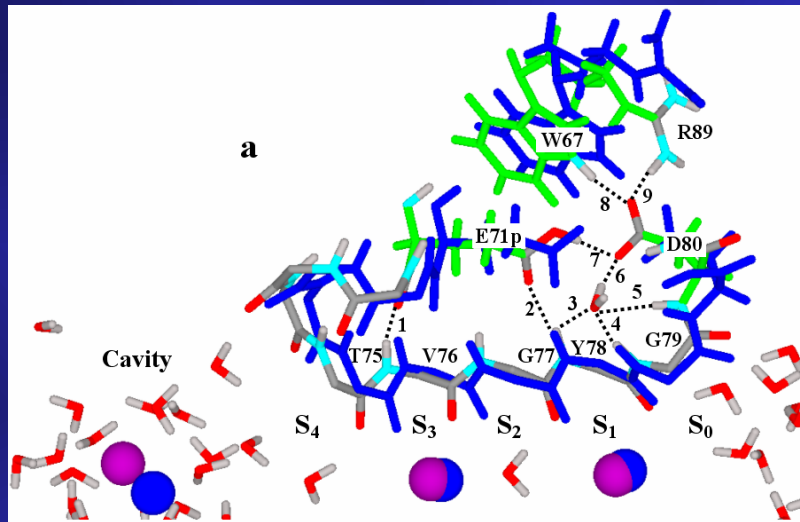
*Telluride, CO*

*July 31, 2007*



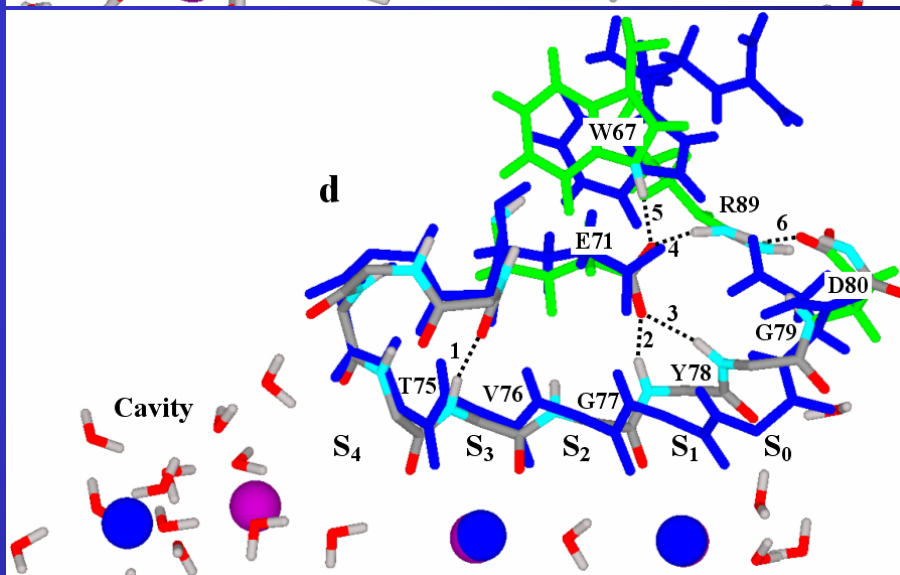
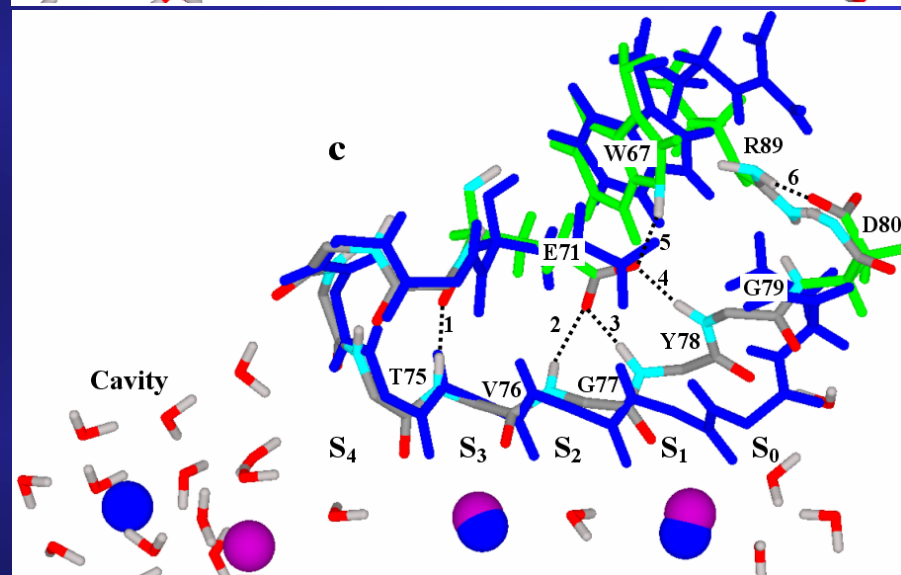
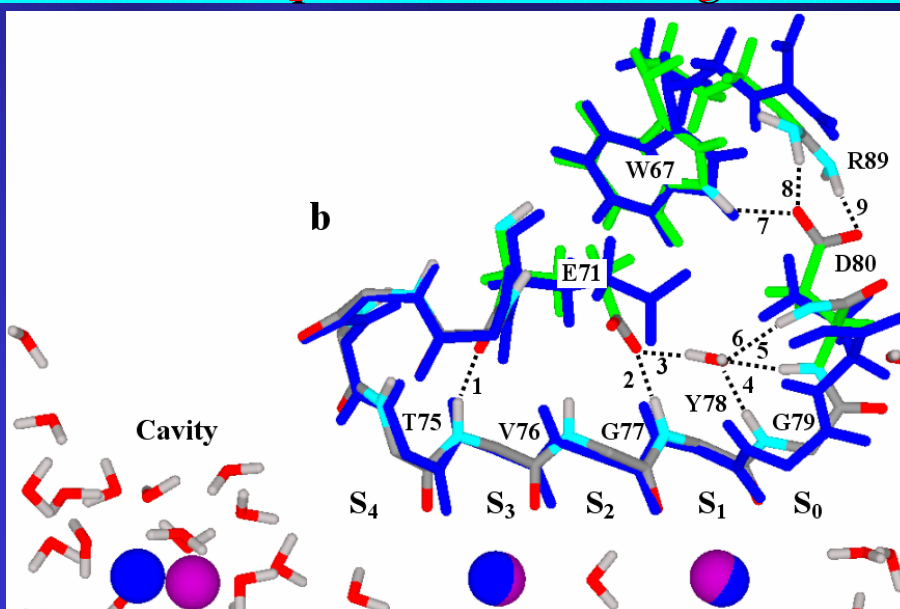
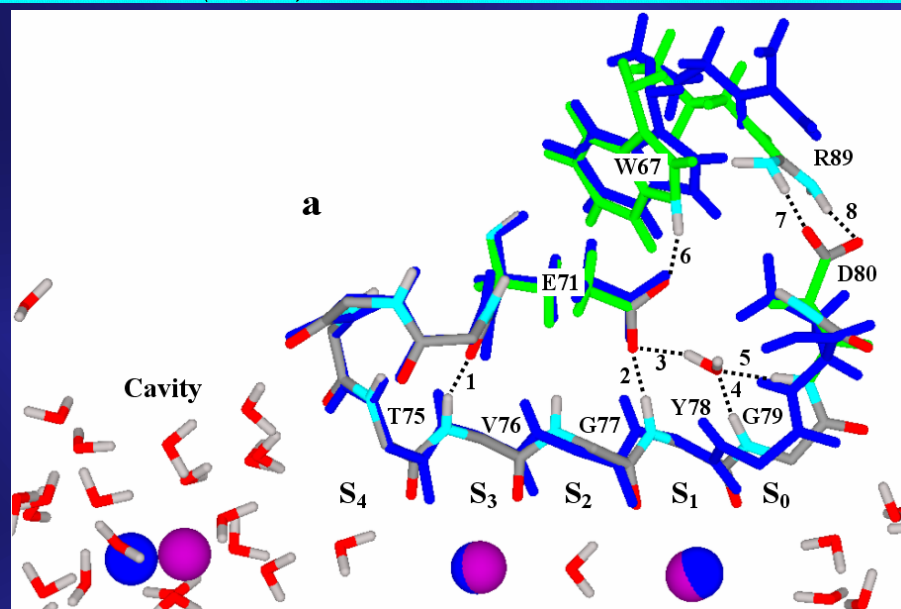
# *Validating E71p KcsA against 1K4C*

*Superposing SF of closed (high [K<sup>+</sup>] structure, 1K4C in blue) and open state (energy-minimized E71p KcsA, native colors) structures. With (a) and without (b) water behind the P-loop*

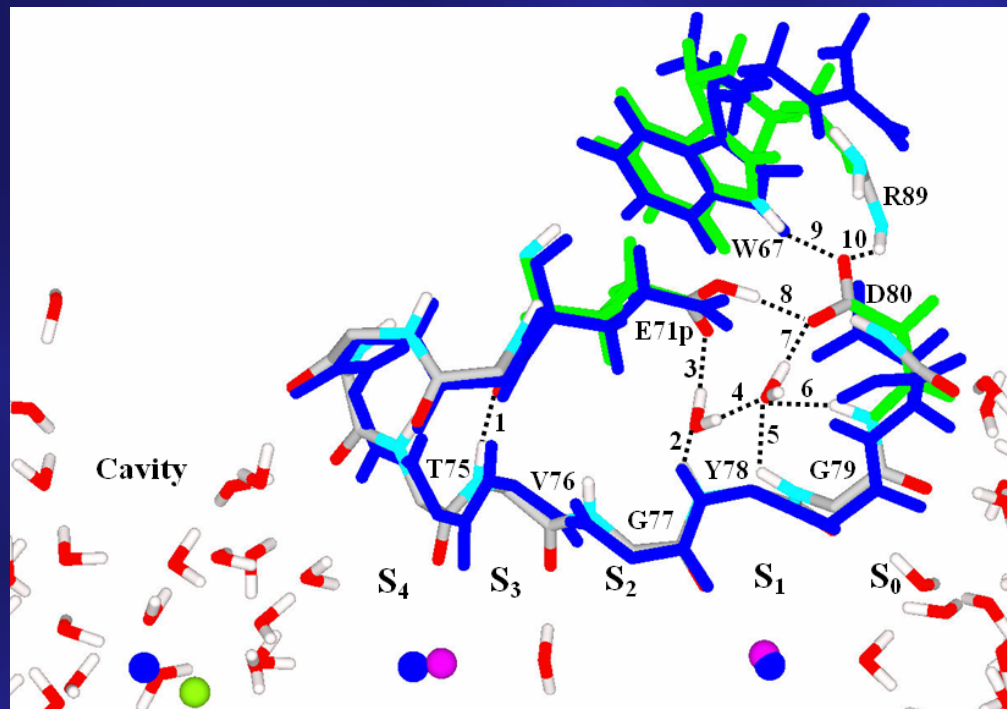


- in both (a) and (b) E71p and D80 stay close to the 1K4C locations, thus strongly supporting the idea that they share a proton
- R89 from the neighboring subunit reorganizes forming a salt bridge with D80
- the amide Hs of both G77 and G79 are not H-bonded in variant (b), but the amide H of G79 is well stabilized by water in variant (a)

*Superposing SF of closed (crystalline 1K4C) and open state (energy-minimized) KcsA structures – E71 and D80 both ionized. With (a,b) and without (c,d) a water molecule behind the P-loop. Note rearrangement.*



# *Validating E71p KcsA against 1K4D*



*Superposing SF of closed (low  $[K^+]$  structure, 1K4D in blue) and open state (energy-minimized E71p KcsA, native colors) structure with a vacant  $S_2$  site and two water molecules behind the P-loop*

➤ E71p and D80 remain close to 1K4D locations

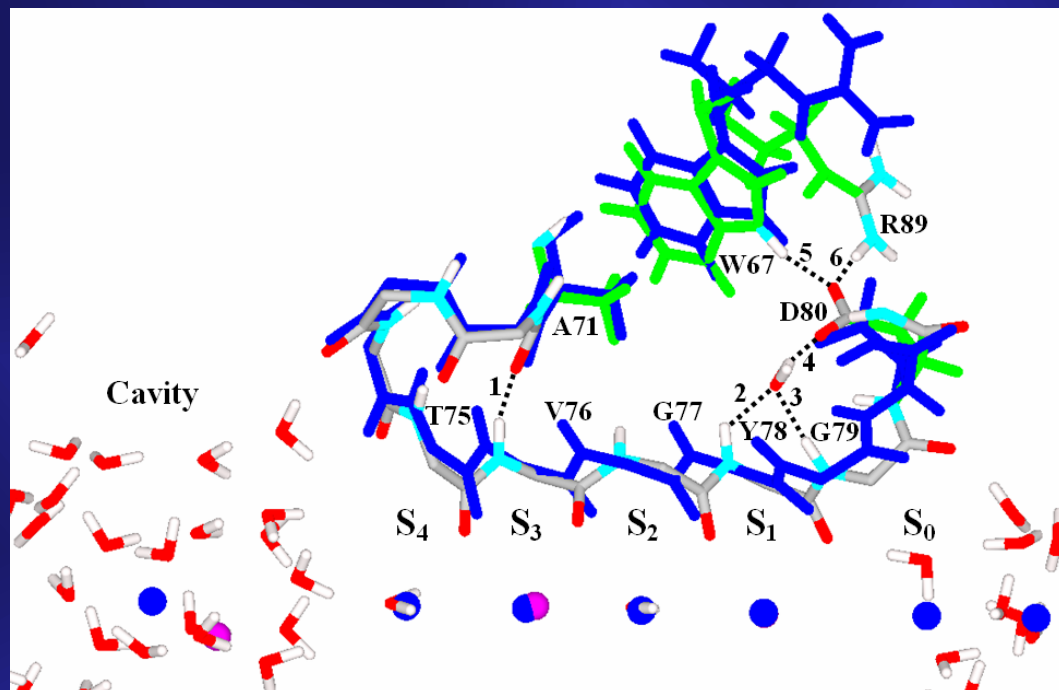
➤ R89 side chain moves toward D80 forming a salt bridge

➤ two water molecules behind the P-loop form H-bonded network with amide Hs of G79 and Y78 and carboxyl Os of E71 and D80

➤ P-loop conformation highly distorted in low  $[K^+]$  structure relative to high  $[K^+]$  structure



# *Validating E71A KcsA against 1ZWI*



*Superposing SF of closed (non-flipped E71A KcsA, 1ZWI in blue) and open state (energy-minimized E71A KcsA, in native colors) structures.*

➤ good overall agreement for P-loops and location of A71, D80 and W67 side chains; R89 side chain moves toward D80 forming a salt bridge

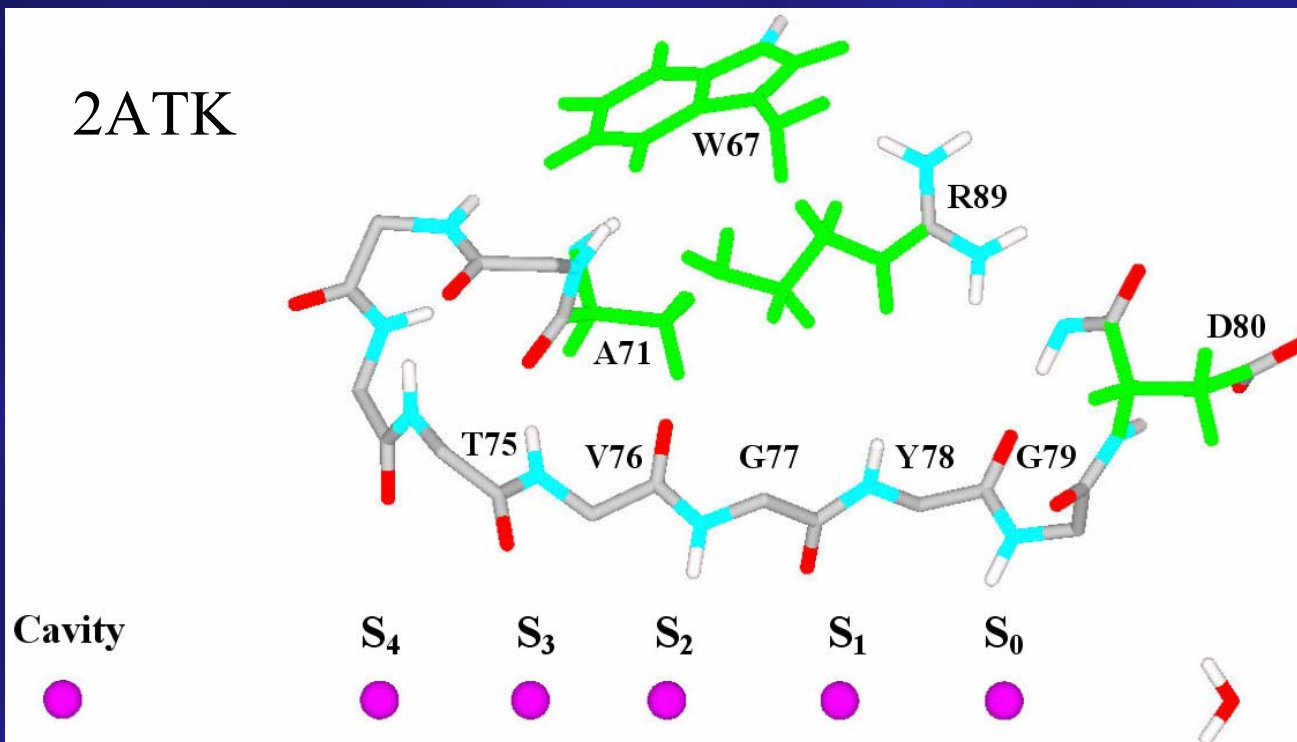
➤ conformation of D80 is close to that in the E71p mutant and differs from that with both E71 and D80 ionized

➤ central part of the P-loop behind the SF is not well stabilized

Cordero-Morales et al. Nat. Struct. Mol. Biol. 13, 311-318 (2006)



# "Flipped" E71A KcsA Crystal Structure



➤ D80 is displaced toward the extra-cellular side; W67 observed in two rotameric states

➤ V76 & Y78 COs flipped away from the pore

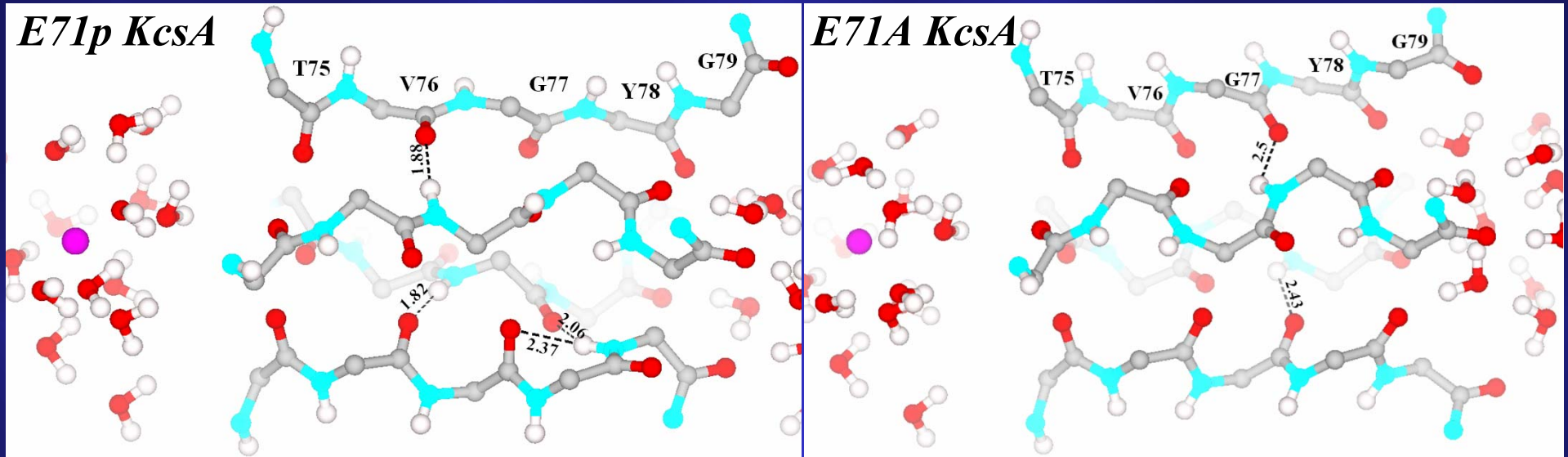
*Conformation of the SF P-loop of the flipped E71A KcsA crystal structure (in native colors), pdb code 2ATK.*

➤ positions of the binding sites are displaced

energy-minimization runs failed to yield conformations similar to this flipped E71A mutant

Cordero-Morales et al. *Nat. Struct. Mol. Biol.* **13**, 311-318 (2006)

# *Conformation of the empty SF*



➤ KcsA remains assembled, but peptide planes of the P-loops undergo strong distortions and flipping;  $C_{\alpha}$  of residues G77, in the center of the SF, twist inward, occluding the pore

➤ the distorted SF is well stabilized by a network of HBs between Hs and Os from the adjacent P-loops

➤ peptide planes **G79-Y78** undergo deviation of  $\sim 90^{\circ}$ ; **Y78-G77** & **G77-V76** planes flip  $\sim 90^{\circ}$ - $180^{\circ}$ ; **V76-T75** plane never strongly distorted

## *Summary*

- energy-minimized open state conformations of E71 mutants agree well with available crystal data except the “flipped” E71A structure
- H-bonding network stabilizing amide Hs behind the P-loops is sensitive to E71 mutations and to the presence of water molecule(s)
- peptide plane distortion and flipping was observed in a SF partially or entirely depleted of ions and waters; the SF void of ions and waters undergoes a conformational change involving all four P-loops and the new conformation is well stabilized by H-bonds between amide Hs and COs from the adjacent P-loops
- our data suggest that inactivated state of the SF corresponds to conformations with partially unoccupied or entirely empty SF
- the selectivity filter is very flexible, and can easily rearrange depending on the supply of ions or the configuration of hydrogen-bonding network stabilizing the P-loops

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# *Current Perspectives on Ion Selectivity*

## *MD Simulations (Sansom's group):*

- $K^+$  ions sit in a cage of eight COs,  $Na^+$  ions interact tightly with a single ring of four COs plus two waters
- the mean  $Na^+$ -O distance is  $\sim 2.36$  Å;  $K^+$ -O distance is  $\sim 2.85$  Å
- SF is blocked by collapse (local constriction of a CO ring) around  $Na^+$  ions

Biggin et al. *BBA* **1510**, 1-9 (2001);      Shrivastava et al. *BJ* **83**, 633-645 (2002)

## *Reduced Models:*

- discrimination against  $Na^+$  due to strong binding, filter constriction, and enhanced energetic and positional fluctuations

Asthagiri & Pratt, *JCP* **125**, 024701 (2006)

- selectivity arises from "external" or "topological" constraints/forces imposed on an ion-coordinated complex by the channel protein

Bostick & Brooks, *PNAS* **104**, 9260-9265 (2007)

# *Current Perspectives on Ion Selectivity*

## *Quantum Chemical Calculations:*

- both *environmental* effects and the protein's ability to provide a proper *coordination number* are determinants of selectivity
- both  $\text{Na}^+$  and  $\text{K}^+$  prefer higher coordination numbers in a low dielectric environment, while lower coordination numbers are found in high dielectric surroundings (bulk water)

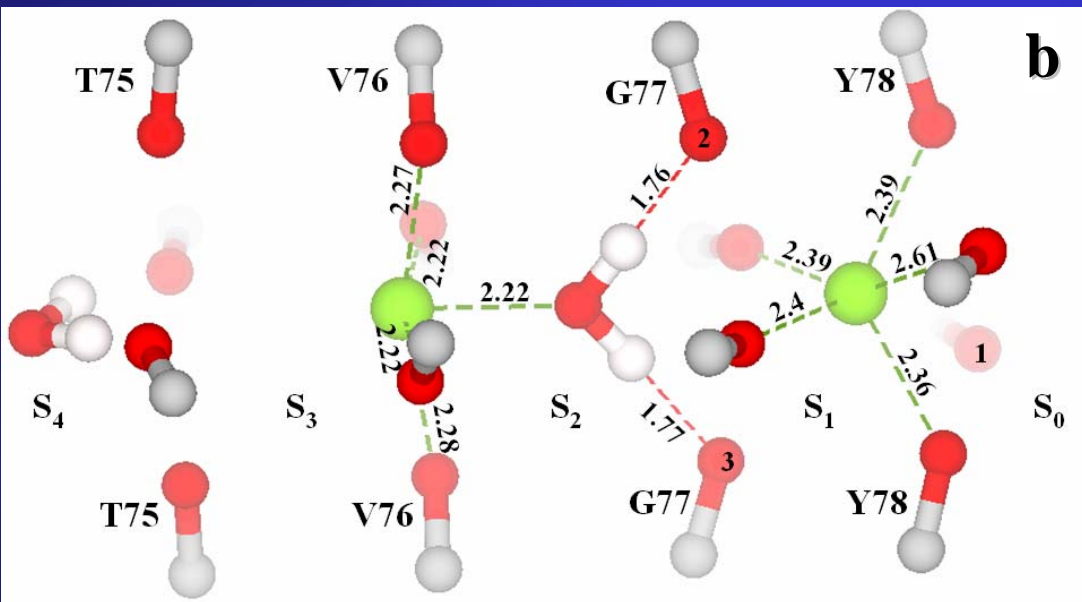
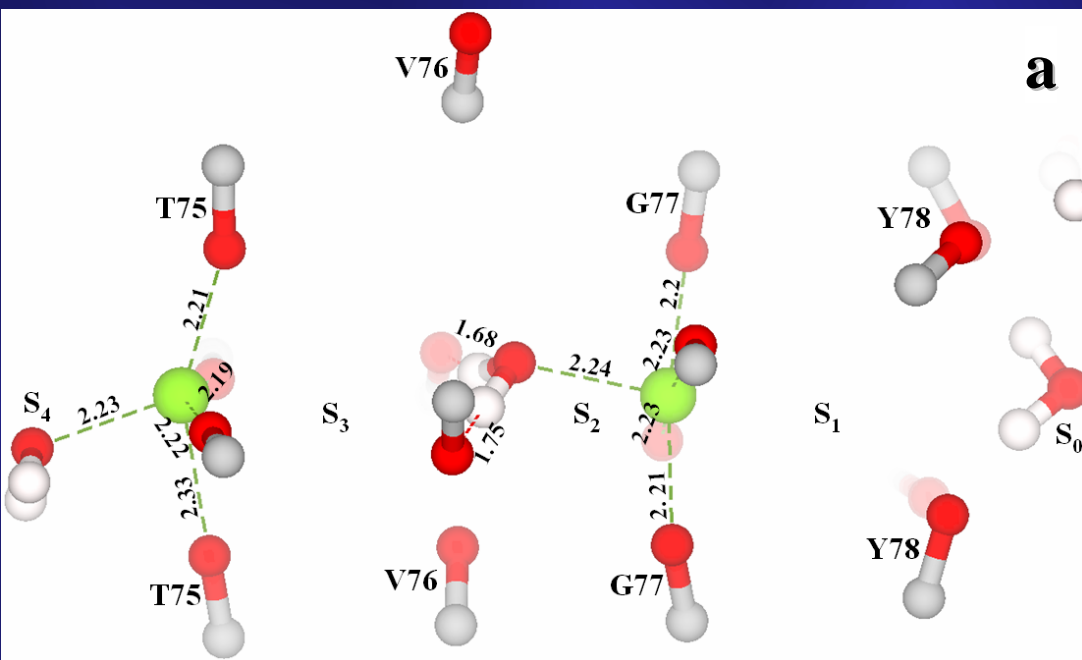
Varma & Rempe, *BJ* **93**, 1093-1099 (2007)

## *MD Simulations (Roux's group):*

- the key to determining selectivity is the strength of the electric field arising from the ligands coordinating the cation
- the physical origin of the strain energy ( $\sim 6$  kcal/mol) in favoring  $\text{K}^+$  over  $\text{Na}^+$  was solely attributed to “through-space” electrostatic repulsions between COs, not to structural deformations of the protein

Noskov et al, *Nature* **431**, 830-834 (2004); Noskov & Roux, *JGP* **129**, 135-143 (2007)

# Conformation of the SF loaded with $Na^+$



The energy-minimized SF loaded with  $\text{Na}^+$  for E71p **(a)** and ionized E71&D80 **(b)** variants. Waters,  $\text{Na}^+$  ions and COs from the four subunits are shown.  $\text{Na}^+$ -O bonds are shown as dashed lines and distances are labeled. In **(a)** one CO of V76 flips. In **(b)**, distance from  $\text{Na}^+$  to Os 1, 2 and 3  $> 2.8 \text{ \AA}$ .

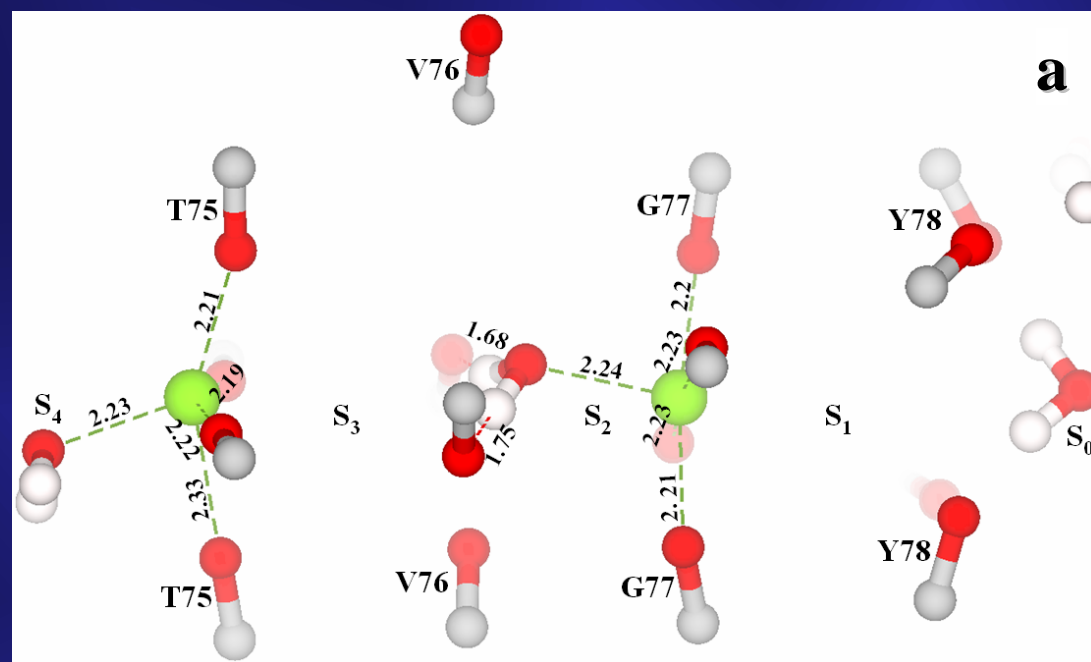
## ***The change of the strain energy***

| <i><b>Energy component</b></i> | <i><b><math>\Delta_{tot}</math> kcal/mol</b></i> | <i><b><math>\Delta_{ion}</math> kcal/mol</b></i> |
|--------------------------------|--|--|
| Electrostatic                  | -65.55   | -21.85   |
| van der Waals                  | 6.41   | 2.14   |
| <b>Total non-bonded</b>        | <b>-59.1</b>                                     | <b>-19.7</b>                                     |
| Bond                           | 1.19   | 0.397  |
| Angle                          | 4.89   | 1.63   |
| Urey-Bradley<br>Angle          | 0.0945   | 0.0315   |
| Dihedral                       | 3.11   | 1.037  |
| Improper                       | 0.419  | 0.1397   |
| <b>Total bonded</b>            | <b>9.7</b>                                       | <b>3.23</b>                                      |
| <b>Total</b>                   | <b>-49.44</b>                                    | <b>-16.48</b>                                    |

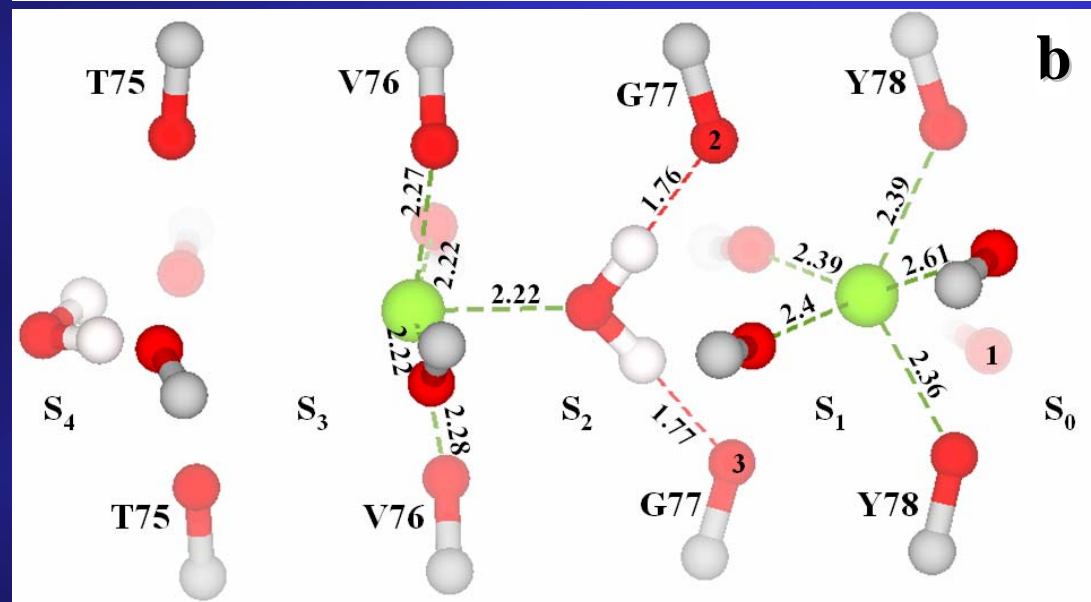
the difference in hydration energy between  $\text{Na}^+$  and  $\text{K}^+$  in bulk solution  $\sim$  **-18** kcal/mol



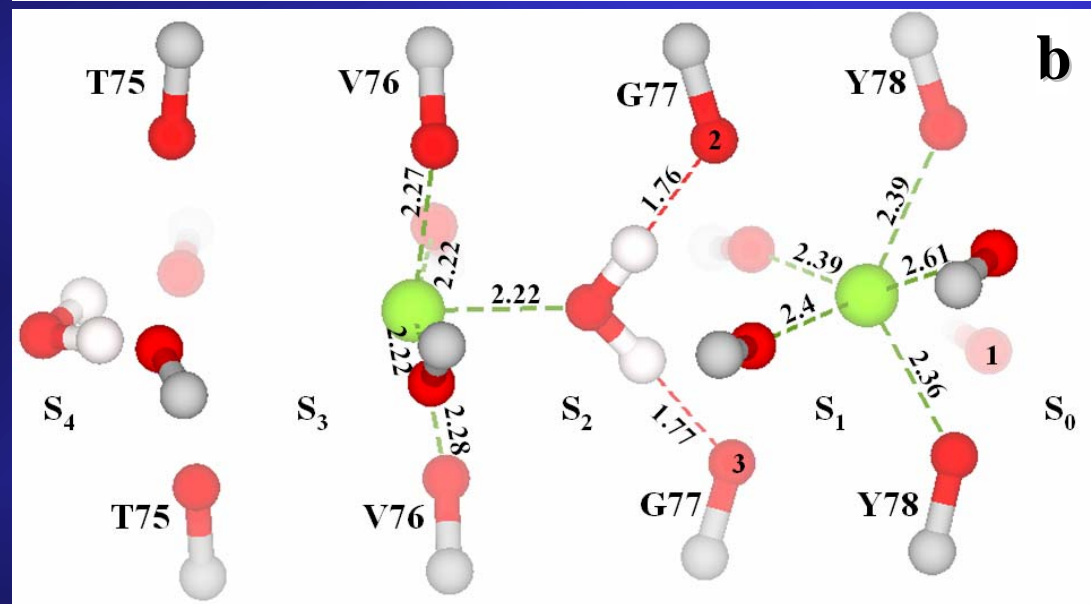
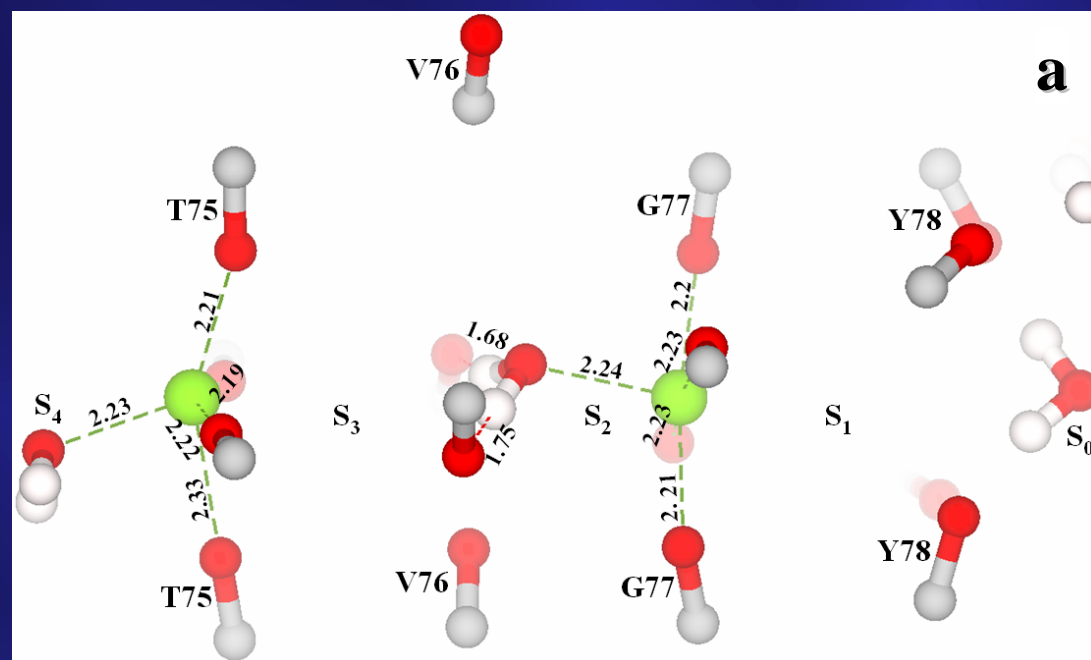
# Conformation of the SF loaded with $\text{Na}^+$



➤ **Conformation change:** the SF rearranges to bind the smaller  $\text{Na}^+$ . The ions either move from the center of S<sub>1</sub> and S<sub>3</sub> toward a ring of COs and are solvated by four COs and a water molecule (a,b) or the original eight-CO cage deforms and five Os collapse around  $\text{Na}^+$  (b). Three COs (labeled 1, 2 and 3 in (b)) do not directly coordinate  $\text{Na}^+$ .

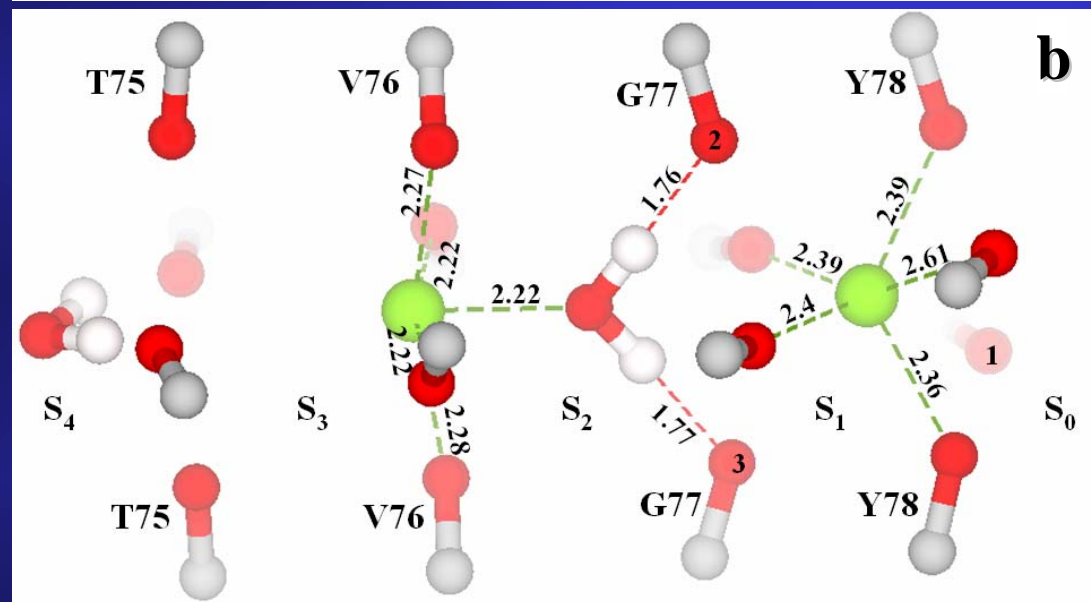
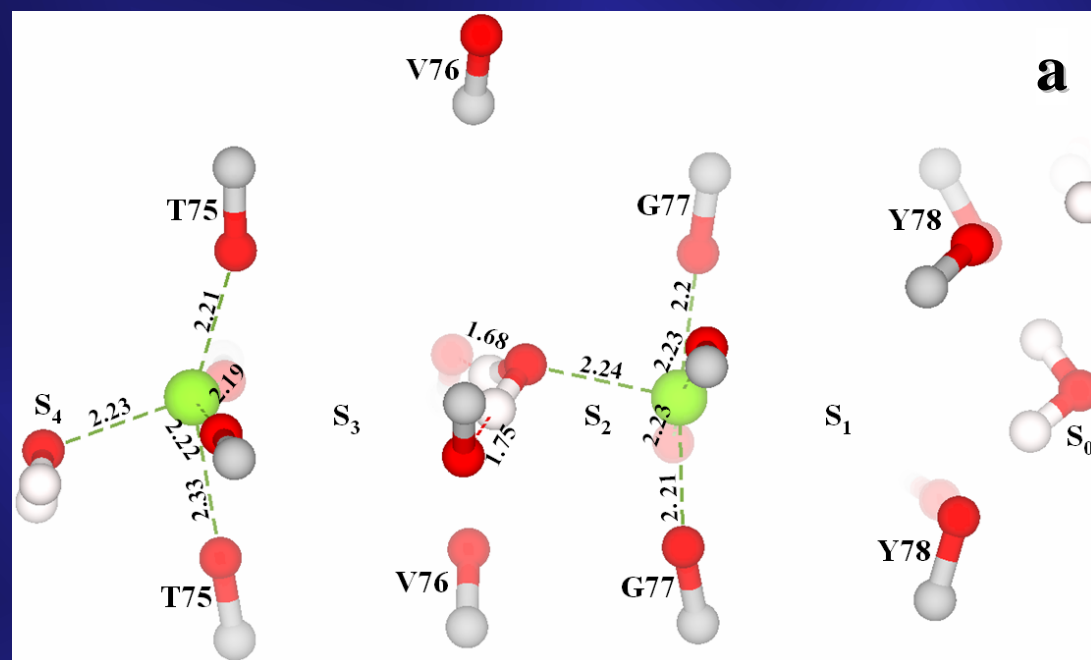


# Conformation of the SF loaded with $\text{Na}^+$



➤ **Coordination number change:** the SF rearranges and five Os properly coordinate  $\text{Na}^+$ ; the SF structure is deformed and presumably nonconductive. Both COs and waters coordinate  $\text{Na}^+$ .

# Conformation of the SF loaded with $\text{Na}^+$



➤ **Size selectivity:** ionic size matters.  $\text{K}^+$  ions fit comfortably in their binding sites without deforming the SF. The  $\text{Na}^+$  binding cavity is smaller; the SF deforms providing five coordinating ligands to favorably cradle  $\text{Na}^+$ . SF collapse around  $\text{Na}^+$  locally occludes the conduction pathway.

# Summary

- **Conformation and binding sites differ with  $\text{Na}^+$  in the SF:**

- The native binding sites (S1-S3) are a good fit for  $\text{K}^+$ , but when  $\text{Na}^+$  replaces  $\text{K}^+$ , these sites are no longer favorable
- The SF rearranges to form new binding sites that comfortably accommodate  $\text{Na}^+$ , with direct coordination to five Os
- Binding is preferably with a ring of four COs and one water molecule, but binding in a deformed eight-fold CO cavity is also seen

- **The SF narrows in the vicinity of  $\text{Na}^+$  ions:**

- The cylindrical geometry of the SF, nearly ideal for  $\text{K}^+$ , is distorted with local contraction around  $\text{Na}^+$ , releasing ligands not involved in  $\text{Na}^+$  coordination and leading to occlusion of the permeation pathway
- Strong binding to  $\text{Na}^+$  induces local collapse of the filter



# Acknowledgements

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