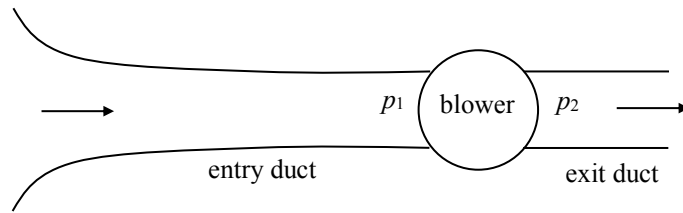


An air blower takes air from the atmosphere (100 kPa and 293 K) and ingests it through a smooth entry duct so that the losses are negligible. The cross-sectional area of the entry duct just upstream of the blower and that of the exit duct are both  $0.01 \text{ m}^2$ .

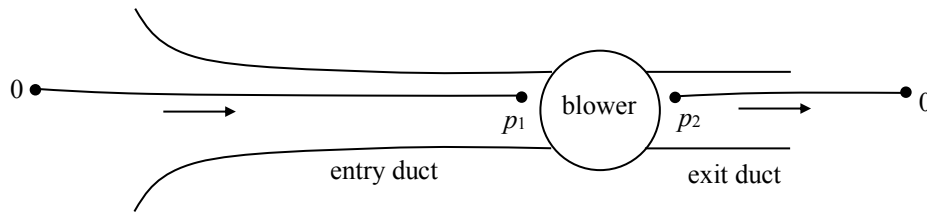


The pressure ratio,  $p_2/p_1$ , across the blower is 1.05 and the exit pressure is equal to atmospheric pressure. The air is assumed to behave isentropically upstream of the blower. Find:

- the velocity of the air entering the blower, and
- the mass flow rate of air through the system.

SOLUTION:

Apply conservation of energy between points 0 and 1 (refer to the figure below). Assume 1D, steady, isentropic ( $\Rightarrow$  adiabatic) flow. Also neglect potential energy changes since a gas is the working fluid.



$$\dot{m}_1 \left( h + \frac{1}{2} V^2 \right)_1 - \dot{m}_0 \left( h + \frac{1}{2} V^2 \right)_0 = 0 \quad (1)$$

The velocity far upstream is negligible ( $V_0 \approx 0$ ) and  $\dot{m}_1 = \dot{m}_0$ . Substitute and simplify.

$$h_0 = h_1 + \frac{1}{2} V_1^2 \quad (2)$$

Assume that the air behaves as a perfect gas ( $\Delta h = c_p \Delta T$ ) and solve for  $V_1$ .

$$V_1 = \sqrt{2c_p (T_0 - T_1)} \quad (3)$$

Express the temperature at point 1 in terms of a pressure ratio making use of the fact that the flow is isentropic.

$$\frac{T_1}{T_0} = \left( \frac{p_1}{p_0} \right)^{\frac{\gamma-1}{\gamma}} \quad (4)$$

The pressure rise across the blower is specified in the problem statement. Furthermore, the pressure at point 2 is equal to atmospheric pressure, i.e.,  $p_2 = p_0$ . Re-write Eq. (4) using this information.

$$\frac{T_1}{T_0} = \left( \frac{p_1}{p_2} \right)^{\frac{\gamma-1}{\gamma}} = \left( \frac{p_2}{p_1} \right)^{\frac{1-\gamma}{\gamma}} \quad (5)$$

Combine Eqs. (3) and (5).

$$V_1 = \sqrt{2c_p (T_0 - T_1)} = \sqrt{2c_p T_0 \left( 1 - \frac{T_1}{T_0} \right)}$$

$$\therefore V_1 = \sqrt{2c_p T_0 \left[ 1 - \left( \frac{p_2}{p_1} \right)^{\frac{1-\gamma}{\gamma}} \right]} \quad (6)$$

Using the given numerical data:

$$c_p = 1004.5 \text{ J/(kg}\cdot\text{K)} \text{ (for air)}$$

$$\gamma = 1.4 \text{ (for air)}$$

$$T_0 = 293 \text{ K}$$

$$p_2/p_1 = 1.05$$

$$\Rightarrow \boxed{V_1 = 90 \text{ m/s}}$$

The mass flow rate can be found from the conditions at point 1:

$$\dot{m} = \rho_1 V_1 A_1 = \left( \frac{p_1}{RT_1} \right) V_1 A_1 \quad (7)$$

where the ideal gas law has been used. The pressure at point 1 can be found from the blower pressure ratio and the fact that  $p_2$  is equal to atmospheric pressure.

$$p_1 = \left( \frac{p_1}{p_2} \right) p_0 \quad (8)$$

The temperature at point 1 can be found using Eq. (5). Using the given numerical data:

$$\begin{aligned} p_0 &= 100 \text{ kPa} \\ p_2/p_1 &= 1.05 \\ p_1 &= 95.2 \text{ kPa} \\ T_0 &= 293 \text{ K} \\ T_1 &= 289 \text{ K} \\ R &= 287 \text{ J/(kg}\cdot\text{K)} \text{ (for air)} \\ V_1 &= 90 \text{ m/s} \\ A_1 &= 0.01 \text{ m}^2 \\ \Rightarrow \dot{m} &= 1.03 \text{ kg/s} \end{aligned}$$