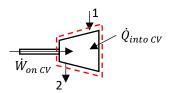
An air compressor operates at steady-state with air entering at  $p_1$ = 15 psfa and  $T_1$ = 60 °F. The air undergoes a polytropic process and exits at  $p_2$  = 75 psfa and  $T_2$  = 294 °F.

- a. Evaluate the exponent for the polytropic process.
- b. Sketch the process on a *p-v* diagram.

## SOLUTION:



Since the process is polytropic,

$$p_1 v_1^n = c = p_2 v_2^n \implies \frac{p_2}{v_1} = \left(\frac{v_1}{v_2}\right)^n,$$
 (1)

$$pv = RT \implies \frac{v_1}{v_2} = \frac{T_1 p_2}{T_2 n_1},\tag{2}$$

Since the process is polytropic, 
$$p_1v_1^n=c=p_2v_2^n\Rightarrow\frac{p_2}{p_1}=\left(\frac{v_1}{v_2}\right)^n, \tag{1}$$
 where  $c$  is a constant and  $n$  is currently unknown. With the air treated as an ideal gas, 
$$pv=RT\Rightarrow\frac{v_1}{v_2}=\frac{T_1p_2}{T_2p_1}, \tag{2}$$
 and combining with Eq. (1), 
$$\frac{p_2}{p_1}=\left(\frac{T_1p_2}{T_2p_1}\right)^n\Rightarrow\ln\left(\frac{p_2}{p_1}\right)=n\ln\left(\frac{T_1p_2}{T_2p_1}\right)\Rightarrow \boxed{n=1.30}, \tag{3}$$
 using the given values of, 
$$p_1=15 \text{ psfa}, \ p_2=75 \text{ psfa}, \ T_1=60 \text{ °F}=519.67 \text{ °R}, \ T_2=294 \text{ °F}=753.67 \text{ °R}.$$

 $p_1 = 15$  psfa,  $p_2 = 75$  psfa,  $T_1 = 60$  °F = 519.67 °R,  $T_2 = 294$  °F = 753.67 °R.

The specific volumes at the inlet and outlet may be found using Eq. (2) with  $R_{air} = 53.3533$  ft.lb<sub>f</sub>/(lb<sub>m</sub>. °R),  $v_1 = 1850 \text{ ft}^3/\text{lb}_m \text{ and } v_2 = 536 \text{ ft}^3/\text{lb}_m.$ 

